GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021(COGC-2021) Semester-V

Course Title: Mass Transfer - II (Course Code: 4350501)

Diploma Programme in which this course is offered	Semester in which offered
Chemical Engineering	5 th Semester

1. RATIONALE

Diploma Chemical engineer have to supervise the preliminary purification of raw materials or final separation of products from by-products. They have to deal with changes in composition of solutions known as the mass-transfer operations. The large numbers of towers used for petroleum refining are examples of mass transfer operations. A substantial number of the unit operations of chemical engineering are concerned with the problem of changing the compositions of solutions and mixtures through methods involving chemical reactions. Hence the course has been designed to develop these competencies and its associated cognitive, practical and effective domain learning out comes.

2. COMPETENCY

The course should be taught and curriculum should be implemented with the aim to develop required skills so that students are able to acquire following competency:

• Perform separation operations for purification of raw materials and products

3. COURSE OUTCOMES(COs)

The theory, practical experiences and relevant soft skills associated with this course are to be taught and implemented, so that the student demonstrates the following industry oriented COs associated with the above mentioned competency:

CO-1: Operate equipment for various gas liquid contacting operations.

CO-2: Apply concept of distillation to various process industries.

CO-3: Use concept of humidification to various process industries.

CO-4: Apply concept of Absorption in Process Industries

CO-5: Use of Ion exchange and Adsorption in Chemical Industries

CO-6: Apply concept of crystallization in process industries

4. TEACHING AND EXAMINATION SCHEME

Teach	ning So	heme	Total Credits		Exan	nination Sc	heme	
	n Hour		(L+T+P)	Theory Marks		Theory Marks Practical Marks		Total Marks
L	Т	Р	C	CA	ESE	CA	ESE	
3	-	4	5	30*	70	50	50	200

(*):Out of 30 marks under the theory CA,10 marks are for assessment of the micro-project to facilitate the integration of COs, and the remaining 20 marks are the average of 2 tests to be taken during the semester for assessing the attainment of the cognitive domain UOs

required for the attainment of the COs.

Legends: L-Lecture; **T** – Tutorial/Teacher Guided Theory Practice; **P** - Practical; **C** – Credit, **CA** -Continuous Assessment; **ESE**-End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

The following practical outcomes (PrOs) are the sub-components of the COs. Some of the **PrOs** marked '*' (in approx. Hrs column) are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

Sr. No.	Practical/Exercise (Course Outcomes in Psychomotor Domain according to NBA Terminology)	Unit No.	Approx. Hrs Required
1	Demonstrate principle, construction and working of equipments for gas–liquid operations with models		4
2	Prepare vapour liquid equilibria curve at atmospheric pressure for Benzene-Xylene		
3	Carry out simple distillation in glass assembly		4
4	Find out the effect of vacuum on distillation of liquid		4
5	Carry out continuous rectification in packed column		
6	Find out amount of steam required in steam distillation		4
7	Find out the property of atmospheric air with the help of wet bulb and dry bulb temperature		4
8	Set desired conditions of humid air in humidity control cabin		4
9	Find out rate of absorption in a tray tower		4
10	Find out rate of absorption in a packed tower		4
11	Characterize industrial adsorbents and observe their samples 4		
12	Remove colour impurities from water using charcoal 4		
13	Find out the yield of crystals from saturated solution without seeding 4		
14	Find out the yield of crystals of from saturated solution with seeding		4
Tota			56

<u>Note</u>

- i. More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.
- ii. The following are some sample 'Process' and '#Product' related skills (more may be added/deleted depending on the course) that occur in the above listed Practical Exercises of this course required which are embedded in the COs and ultimately the competency.

Sr.No.	Sample Performance Indicators for the PrOs	Weight age in %
1	Handling of apparatus for precise measurements	10
2	Record observations correctly	20
3	Practice and adapt good and safe measuring techniques	10
4	Calculations, Interpretation of results and their conclusion.	20

5	Prepare report of practical in prescribed format	10
6	Solve assignment questions.	20
7	Viva-voce	10
Tota	1	100

6. MAJOR EQUIPMENT/INSTRUMENTS AND SOFTWARE REQUIRED

These major equipment/instruments and Software required to develop PrOs are given below with broad specifications to facilitate procurement to them by the administrators/management to the institutes. This will ensure the conduction of practice in all institutions across the state in a proper ways other the desired skills are developing in students.

- Distillation Assembly: 2000 ml round bottom flask, 1000 ml collection flask, joints, adapter with ¾ neck, simple/coiled glass condenser, thermometer pocket
- Steam distillation setup: Distillation kettle MOC-MS, dia-150 mm, height 300mm; jacket dia 175 mm height, height 300 mm, pressure gauge, steam relief valve, steam feed line with valve, drain valve, steam trap on jacket outlet, 25 mm glass wool insulation with MS cladding; Condenser – MS shell, tube copper dia-150 mm, height 250; Steam generator inner SS 304, outer MS dia 180 mm, height 270 mm; 25,5litre collecting beaker
- 3. VLE apparatus: Heating mantle with 1-liter flask, dimmer stat, digital temp indicator, air- and water-cooled condenser, mounted on wooden and MS frame, thermocouples
- 4. Humidity cabin: Double walled thick gauge chamber SS 304, heater 500 W; Cooling circuit with compressor, expansion valve, condenser and refrigerant; Steam generator SS 304; Control panel with digital temperature indicator, low water level indicator, solenoid valve
- Batch crystallizer: Jacket 325 mm round, 155 mm deep, 3mm thick, annulus 22.5 mm; 25 mm thick glass wool insulation, Aluminum cladding; motorstirrer 10mm rod, speed regulator
- 6. Benzene, Toluene, Xylene, Sand, Limestone, silica gel, Charcoal, boric acid, Sodium sulphate, Potassium permanganate

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs and PrOs. More could be added to fulfill the development of this competency.

- a) Work as a leader/a team member.
- b) Follow ethical practices
- c) Observe safety measures
- d) Good house keeping
- e) Time management
- f) Practice environmentally friendly methods and processes.

The ADOs are best developed through laboratory/field-based exercises. Moreover, the level

of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year
- ii. 'Organization Level' in 2nd year.
- iii. 'Characterization Level' in 3rd year.

8. UNDERPINNING THEORY

The major underpinning theory is given below based on the higher level UOs of *Revised Bloom's taxonomy* that are formulated for development of the COs and competency. If required, more such higher-level UOs could be included by the course teacher to focus on the attainment of COs and competency.

Unit	Major Learning Outcomes	Topics and Sub-topics
	(in cognitive domain) 1a Describe importance	1.1 Importance of Gas-Liquid
	of Gas-Liquid operations 1b Classify equipments for	operations 1.2 Classification operations of equipments for Gas-Liquid
Unit I Equipment for Gas Liquid Operations	Gas Liquid operations 1c Describe construction of equipments with diagram of 1.3 &1.4 1d Explain working principle and operation of equipments with sketches of 1.3 &1.	 1.3 Gas dispersed, Mechanically Agitated Vessel, Tray tower, Types of trays, Operating problems in tray tower, Real Tray & Tray efficiency- point efficiency, Murphy efficiency, Overall-Tray efficiency (only definition) 1.4 Liquid dispersed, Spray tower, Packed tower and its operating problems
	1e Distinguish different packing with diagram types of	1.5 Types of packing (a) Random (b) Regular
	2a Describe applications	2.1 Distillation as a versatile separation method
	2b Describe the steps to Plot VLE, Constant pressure, Constant temperature equilibria	2.2 Vapor Liquid Equilibria, Constant pressure equilibria and Constant temperature equilibria
	2c Explain Relative volatility and laws - Raoult's, Henry's 2c.1State their uses	2.3 Relative volatility2.4 Raoult's law, Henry's law, and their uses
	2d Differentiate azeotropes	2.5 Maximum and minimum boiling azeotropes
Unit II Distillation	2eExplain-Flashvaporization,Differentialdistillation,Continuous rectification2fCalculateamountandcompositionforFlashvaporization2gCalculateproduct	 2.6 Flash vaporization, Material balance and Calculation of amount and composition 2.7 Differential distillation, Derivation of Rayleigh's equation and
	composition for Differential distillation	Calculation of product composition 2.8 Continuous rectification of binary solution, The fractionation operation and Overall material balances
	 2h Apply McCabe-Theile method for multistage tray tower for enriching and stripping section 2i Calculate product rates, minimum reflux ratio and number of trays for the given data 	2.9 McCabe and Thiele method for enriching and stripping section, Introduction of Feed and Location of the feed tray, Total reflux ratio, Minimum reflux ratio, Optimum reflux ratio, calculations of product rates, minimum reflux

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	Ι	ratio and number of trays
	2j Compare distillation techniques viz (a) Steam distillation, (b) Vacuum and molecular distillation(c) Azeotropic and extractive distillation	2.10 Important distillation technique Steam distillation, Vacuum and molecular distillation, Azeotropic and extractive distillation
	2k Distinguish Reboilers	2.11 Reboilers and their use
	3.a Analyse the VLE for a pure substance	3.1 Vapor-pressure curve3.2 Saturated and unsaturated vapor- gas mixtures
Unit III	3.b Explain the concepts of Absolute humidity, Relative saturation, Percentage saturation, Dew point, Dry bulb temperature, Wet bulb temperature, Adiabatic saturation temperature, Humid volume, Humid heat, Enthalpy	3.3 Concept of Absolute humidity, Relative saturation, Percentage saturation, Dew point, Dry bulb temperature, Wet bulb temperature, Adiabatic saturation temperature, Humid volume, Humid heat, Enthalpy
Humidification	3.c Evaluate the property of air using DBT and WBT Calculate – absolute humidity, relative saturation, percentage saturation for the given process data	3.4 Calculations of absolute humidity, relative saturation, percentage saturation
	3.d Draw psychometric chart List Purposes of contact of gas with pure Liquid3.e Explain construction and	 3.5 Psychometric charts for Air-Water system 3.6 Purposes of contact of gas with pure Liquid 3.7 Equipments, Cooling towers.
	working with diagram	
	4.a Apply concept of absorption4.b Describe the physical properties of gases4.c Explain Raoult's law	 4.1 Industrial application of Absorption Gas Absorption 4.2 Equilibrium solubility of gases in liquids and effect of temperature and pressure.
	A d Salact appropriate solvent	4.3 Ideal solution and Raoult's law4.4 Solvent for absorption
Unit IV Gas Absorption	 4.d Select appropriate solvent 4.e Explain Material balance in different condition 4.f Select liquid-gas ratio for absorber 	 4.5 Material balance for one component transfer 1. Counter current flow 2. Co-current flow 3. counter current multistage operation 4.6 Minimum liquid-gas ratio for absorber
	4.g Explain tray tower and packed tower4.h Evaluate various packing	4.7 HETP

	4.i Calculate absorption based on material balance	4.8 Raoult's law and material balance applied in gas absorption
	5.a Define and state uses of Adsorption	5.1 Definition and industrial application of Adsorption
	5.b Classify Adsorption and adsorbents5.c State Commonly used adsorbents	5.2 Types of adsorptions, Nature of adsorbents, commonly used adsorbents
Unit V	5.d Analyse Adsorption Equilibria 5.e Describe Effect of temperature on adsorption and Heat of adsorption	5.3 Adsorption Equilibria, Adsorption hysteresis, Effect of temperature on adsorption and Heat of adsorption
Adsorption and Ion Exchange	5.f Apply Freudlich's equation for single stage and multi stage cross-current operation5.g Describe adsorption from dilute and concentrated solution	5.4 Adsorption from liquids, Adsorption from dilute solution, The Freundlich's equation, Adsorption from concentrated solutions, Material balance and Freundlich's equation for single stage operation.
	5.h Describe construction and working of Higgins contactor, Pressure swing adsorber	5.5 Higgins contactor and Pressure swing adsorber
	5.i Appreciate concepts of IonExchange5.j List Application of Ion Exchange	5.6 Ion-Exchange: Principles, Application.
	6.a State Industrial applications of crystallization	6.1 Industrial applications of crystallization
	6.b Explain equilibria mechanismfor crystallization6.c State the methods to get Supersaturation	6.2 Equilibria and yields, Super saturation and methods to get it, Nucleation and Crystal growth
Unit VI Crystallization	 6.d Explain working principle and operation of Crystallization Equipment with sketch 6.e Describe construction of Crystallization Equipment 	6.3 Crystallization Equipment, Vacuum crystallizer, Swenson walker crystallizer.
	6.f State and explain Meir's theory 6.g Calculate the crystal yield	 6.4 Meir's theory 6.5 Crystallization with and without seeding 6.6 Calculations of crystal yield
	6.h List steps to Prevent caking of crystals	6.6 Calculations of crystal yield 6.7 Caking of crystals and its prevention

9. SUGGESTED SPECIFICATION TABLE FOR QUESTION PAPER DESIGN

Unit Unit Title Distribution of Theory Marks
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		Teaching Hours	R Level	U Level	A Level	Total Marks
I	EquipmentforGas LiquidOperations	04	2	03	02	7
11	Distillation	12	05	06	08	19
111	Humidification	06	02	04	03	09
IV	Gas Absorption	07	02	04	08	14
V	Adsorption&lon- Exchange	08	04	04	04	12
VI	Crystallization	05	03	03	03	09
Total		42	18	24	28	70

Legends: R=Remember, U=Understand, A=Apply and above (Revised Bloom's taxonomy) <u>Note</u>: This specification table provides general guidelines to assist students for their learning and to teachers to teach and question paper designers/setters to formulate test items/questions to assess the attainment of the UOs. The actual distribution of marks at different taxonomy levels (of R, U and A) in the question paper may slightly vary from above table.

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, following are the suggested student-related *co-curricular* activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform following activities in group and prepare reports of about 5 pages for each activity. They should also collect/record physical evidences for their (student's) portfolio which may be useful for their placement interviews:

Following is the list of proposed student activities like:

- 1. Assignments
- 2. Technical Quiz/MCQ Test
- 3. Presentation on some course topic
- 4. I-net based assignments
- 5. Undertake micro-Project in team/individually

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Massive open online courses (*MOOCs*) may be used to teach various topics/subtopics.
- b) Guide student(s) in undertaking micro-projects/activities.
- c) Different types of teaching methods i.e. video demonstration, activity based learning, case study, m-learning need to be employed by teachers to develop the outcomes.
- d) Some *of the topics/sub-topics* which is relatively simpler or descriptive is to be given to the students for *self-learning* but to be assessed using different assessment methods.
- e) Teachers need to ensure to create opportunities and provisions for *co-curricular activities*.
- f) Guide students to address issues on environment and sustainability with reference to using the knowledge of this course

g) OERs, Vlab, and Olabs may be used to teach for the teaching of different concepts.

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her at the beginning of the semester. In the first four semesters, the micro-project is group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application-based, internet-based, workshop-based, laboratory-based, or field-based. Each micro-project should encompass two or more COs which are the integration of PrOs, UOs, and ADOs. Each student will have to maintain dated work diary consist in go find individual contributions in the project work and give a seminar presentation of it before submission. The duration of the micro project should be about **14**-**16** *(fourteen to sixteen) student engagement hours* during the course. The students ought to submit micro-project by the end of the semester (so that they develop industry-oriented COs).

A suggestive list of micro-projects is given here. This should relate highly to the competency of the course and the COs. Similar micro-projects could be added by the concerned course teacher.

1.	Prepare chart/model of Mass transfer operation and applications.
2.	Prepare 15-20 slides power point presentation onmass transfer operation along with their examples.
3.	Prepare 15-20 slides power point presentation on topic of mass transfer operation.
4.	Prepare Laboratory set up for distillation, absorption, crystallization, humidification, Gas liquid operation or ion exchange.
5.	Prepare a demonstrative model of any mass transfer equipment.
6.	Prepare Working model of any mass transfer equipment.

13. SUGGESTED LEARNING RESOURCES

Sr. No.	Title of Books	Author	Publication with place, year and ISBN	
1	Mass Transfer Operations	Robert E.Treybal	Mc Graw- Hill, 3rd Edition,1981	
2	Unit Operation of Chemical Engineering	McCabe, Warren	McGraw Hill Publication, NewYork2004, 7th Edition	
3	Unit Operations-II	K.A. Gavhane	Nirali Prakashan, Pune	
4	Unit Operations of Chemical Engineering,Volume-I	P. Chattopadhyay	Khanna Publishers, NewDelhi,1995	
5	Chemical Engineering,Volume-2	Coulsion and Richardson	Butterworth-Heinemann; 5 th Edition, 2002	
7	Introduction to Chemical Engineering	L.Badger, Julius T.Banchero	McGraw Hill Publication, NewYork, 7 th Edition, 2004	

14. SUGGESTED LEARNING WEBSITES

- **a.** www.unitoperation.com
- **b.** http://nptel.ac.in/courses/index.php?subjectId=103103035
- **c.** http://1rv07ch.files.wordpress.com/2010/05/lecture1introduction2mass-transfer.pdf
- d. http://www.msubbu.in/ln/mt/
- e. http://chemeng.ir/download/Mass-Transfer/Mass_Transfer_Operations_-Robert_Treybal_chemeng.ir.pdf
- f. http://serve.me.nus.edu.sg/arun/file/teaching/ME6203_2013_Mujumdar.pdf

15. PO-COMPETENCY-CO MAPPING

Semester V	Mass Transfer -II (4350501)							
Semester				POs	5			
Competency & Course Outcomes	PO1 Basic & Discipline -specific knowledg e	PO2 Problem Analysis	PO3 Design/dev elopment of solutions	PO4 Engineerin g Tools, Experiment ation &Testing	PO5 Engineering practices for society, sustainability & environment	PO6 Project Manageme nt	PO7 Life- long learning	
Competency	Supervi	ise opera	tion and m	aintenance	of various Ma	ss Transfer e	quipments	
CO-1: Operate equipment for various gas liquid contacting operations.	2.00	3.00	3.00	2.00	-	1.00	2.00	
CO-2: Apply concept of distillation to various process industries.	2.00	2.00	2.00	2.00	-	-	1.00	
CO-3: Use concept of humidification to various process industries.	2.00	1.00	2.00	1.00	2.00	-	-	
CO-4: Apply concept of Absorption in Process Industries	2.00	2.00	2.00	1.00	2.00	-	-	
CO-5: Use of Ion exchange and Adsorption in Chemical Industries	2.00	3.00	2.00	2.00	2.00	-	1.00	

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CO-6: Apply concept of crystallization in	2.00	2.00	2.00	2.00	2.00	-	1.00
process industries							

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021(COGC-2021) Semester-V

Course Title: Waste to Energy Conversion Technology

(Course Code: 4350502)

Diploma Programme in which this course is offered	Semester in which offered
Chemical Engineering	5 th Semester

1. RATIONALE

The objective of the course is to provide insights into waste management options by reducing the waste destined for disposal and encouraging the use of waste as a resource for alternate energy production. This course is designed to provide an understanding of the various aspects of Waste to Energy. The various sources of waste generation are analyzed with a focus on its potential for energy production. The need for characterization of wastes will be discussed along with the existing norms for waste utilization for alternate energy source. Various Technological options available for the production of energy form waste.

2. COMPETENCY

The course should be taught and curriculum should be implemented with the aim to develop required skills so that students are able to acquire following competency:

• To Supervise operation and to Optimize energy and Other Inventories

3. COURSE OUTCOMES(COs)

The theory, practical experiences and relevant soft skills associated with this course are to be taught and implemented, so that the student demonstrates the following industry oriented COs associated with the above mentioned competency:

a) Identify different wastes as a energy source

- b) Apply concept of Thermo-chemical process for energy conversion
- c) Apply concept of Biochemical process for energy conversion

d) Utilize algal biomass as source of energy production

4. TEACHING AND EXAMINATION SCHEME

Teach	Teaching Scheme		Total Credits	Examination Scheme					
	n Hours		(L+T+P)	Theory Marks		Practical Marks		Total Marks	
L	Т	Р	C	CA	ESE	CA	ESE		
3	-	-	3	30*	70	-	-	100	

(*):Out of 30 marks under the theory CA,10 marks are for assessment of the micro-project to facilitate the integration of COs, and the remaining 20 marks are the average of 2 tests to be taken during the semester for assessing the attainment of the cognitive domain UOs required for the attainment of the COs.

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Legends: L-Lecture; **T** – Tutorial/Teacher Guided Theory Practice; **P** - Practical; **C** – Credit, **CA** -Continuous Assessment; **ESE**-End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

The following practical outcomes (PrOs) are the sub-components of the COs. Some of the **PrOs** marked '*' (in approx. Hrs column) are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

Sr. No.	Practical/Exercise (Course Outcomes in Psychomotor Domain according to NBA Terminology)	Unit No.	Approx. Hrs Required				
	Not Applicable						

<u>Note</u>

- i. More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.
- ii. The following are some sample 'Process' and '#Product' related skills (more may be added/deleted depending on the course) that occur in the above listed Practical Exercises of this course required which are embedded in the COs and ultimately the competency.

Sr. No.	Sr. No. Sample Performance Indicators for the PrOs			
1	Question answer or Writing steps exercise (Assignment)	30		
2	Executing of exercise	30		
3	Result	40		
	Total			

6. MAJOR EQUIPMENT/INSTRUMENTS AND SOFTWARE REQUIRED

These major equipment/instruments and Software required to develop PrOs are given below with broad specifications to facilitate procurement to them by the administrators/management to the institutes. This will ensure the conduction of practice in all institutions across the state in a proper ways other the desired skills are developing in students.

Sr. No.	r. No. Equipment Name with Broad Specifications	
	Not Applicable	

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs and PrOs. More could be added to fulfill the development of this competency.

- a) Work as a leader/a team member.
- b) Follow ethical practices
- c) Observe safety measures

- d) Good house keeping
- e) Time management
- f) Practice environmentally friendly methods and processes.

The ADOs are best developed through laboratory/field-based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year
- ii. 'Organization Level' in 2nd year.
- iii. 'Characterization Level' in 3rd year.

8. UNDERPINNING THEORY

The major underpinning theory is given below based on the higher level UOs of *Revised Bloom's taxonomy* that are formulated for development of the COs and competency. If required, more such higher-level UOs could be included by the course teacher to focus on the attainment of COs and competency.

	Major Learning	
Unit	Outcomes(Course Outcomes in	Topics and Sub-topics
	Cognitive Domain according to	
	NBA terminology)	
	1a. Classification of waste as	1.1 Definition of wastes and their classification
	energy	1.2 Important quality parameters of different
Unit:1		types of wastes
Characterizati	1c. Characterization of	1.3 Wastes suitable for energy production 1.4 Characterization of solid wastes and waste
on of waste	wastes	water
	1d. Explain Technologies for	1.5 Energy from wastes, some scenario
	Waste to Energy Conversion	1.6 Routes for energy production from wastes
		1.7 Define Biochemical Conversion and Thermo- chemical Conversion
	1b. Explain Status of wastes	1.8 Status of waste to Energy conversion in India.
	1d. Solve simple numerical	1.9 Numerical based on Characterization of
	ru. solve simple numerical	
	2a. Explain Mechanism of	wastes 2.1 Definition and scope of
11	combustion	combustion/incineration
Unit:2		2.2 Environmental and health impacts of
Thermo-		incineration
chemical Conversion		2.3 Advantages and Disadvantages of combustion 2.4 Air requirement for combustion
	2b. Classification of combustors	2.5 Type of combustor
		2.6 Energy production from wastes through
		incineration
	2c. Explain Basic chemistry of gasification	2.7 Concept and Advantages of gasification
	2d. Classification of Gasifier	2.8 Types of Gasifier2.9 Energy production through gasification of wastes
	2e. Comparison Combustion	2.10 Comparison between incineration and
	and gasification	gasification

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routes for algal
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mechanisms
eld

	4.10 Upgradation of algal oil to bio diesel using
	homogenous and heterogeneous catalysts

	Unit Title	Taashing	Distribution of Theory Marks				
Unit		Teaching	R	U	Α	Total	
		Hours	Level	Level	Level	Marks	
I	Characterization of waste	6	2	4	2	8	
П	Thermo-chemical Conversion	14	5	10	10	25	
III	Biochemical Conversion	14	5	10	10	25	
IV	Cultivation of algal	8	6	6	0	12	
	Total	42	18	30	22	70	

9. SUGGESTED SPECIFICATION TABLE FOR QUESTION PAPER DESIGN

Legends: R=Remember, U=Understand, A=Apply and above (Revised Bloom's taxonomy) <u>Note</u>: This specification table provides general guidelines to assist students for their learning and to teachers to teach and question paper designers/setters to formulate test items/questions to assess the attainment of the UOs. The actual distribution of marks at different taxonomy levels (of R, U and A) in the question paper may slightly vary from above table.

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, following are the suggested student-related *co-curricular* activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform following activities in group and prepare reports of about 5 pages for each activity. They should also collect/record physical evidences for their (student's) portfolio which may be useful for their placement interviews:

Following is the list of proposed student activities like:

- 1. Assignments
- 2. Technical Quiz/MCQ Test
- 3. Presentation on some course topic
- 4. I-net based assignments
- 5. Undertake micro-Project in team/individually

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Massive open online courses (MOOCs) may be used to teach various topics/subtopics.
- b) Guide student(s) in undertaking micro-projects/activities.
- c) Different types of teaching methods i.e. video demonstration, activity based learning, case study, m-learning need to be employed by teachers to develop the outcomes.
- d) Some *of the topics/sub-topics* which is relatively simpler or descriptive is to be given to the students for *self-learning* but to be assessed using different assessment methods.
- e) Teachers need to ensure to create opportunities and provisions for *co-curricular activities*.

- f) Guide students to address issues on environment and sustainability with reference to using the knowledge of this course
- g) OERs, Vlab, and Olabs may be used to teach for the teaching of different concepts.

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her at the beginning of the semester. In the first four semesters, the micro-project is group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application-based, internet-based, workshop-based, laboratory-based, or field-based. Each micro-project should encompass two or more COs which are the integration of PrOs, UOs, and ADOs. Each student will have to maintain dated work diary consist in go find individual contributions in the project work and give a seminar presentation of it before submission. The duration of the micro project should be about **14**-**16** (*fourteen to sixteen*) *student engagement hours* during the course. The students ought

to submit micro-project by the end of the semester (so that they develop industry-oriented COs).

A suggestive list of micro-projects is given here. This should relate highly to the competency of the course and the COs. Similar micro-projects could be added by the concerned course teacher.

Prepare chart/model of Classification of waste along with their Mechanism and applications
Prepare chart of Status of waste to Energy conversion in India
Prepare chart/model types of Combustor
Draw suitable chart for various Combustor equipment
Prepare 15-20 slides power point presentation showing Classification of waste along with
their examples
Prepare 15-20 slides power point presentation Characterization of wastes
Prepare 15-20 slides power point presentation on Waste to energy technology
Prepare a demonstrative model of any Gasifier reactor
Prepare a demonstrative model of any waste to energy conversion technology equipment
Prepare Working model of any waste to energy conversion technology equipment
Prepare 15-20 slides power point presentation on energy crisis in India and its solution
Prepare a model on pyrolysis system that can convert plastic into gas or oil.
Prepare chart/model of Incineration.
Prepare 15-20 slides power point presentation on comparison of seven different countries
best practices for waste energy recovery with India.

13. SUGGESTED LEARNING RESOURCES

Sr. No.	Title of Books	Author	Publication with place, year and ISBN
1	Waste-to-Energy: Technologies and Project Implementation	Marc J Rogoff Dr and Francois Screve	William Andrew; 2nd edition (15 June 2011)

2	Biogas Technology	Khandelwal K. C. and	Vol. I & II Tata
2	blogas recimology	Mahdi S. S	McGraw Hill Publishing Co. Ltd., 1983
3	Solid Waste Engineering	Vesilind P.A. and	2nd Ed. Cengage India (2016)
		Worrell W. A	2.14 241 001.8480 11414 (2010)
	Thermo-chemical Processing		
4	of Biomass: Conversion into	Robert C. Brown,	John Wiley and Sons, USA (2019)
	Fuels, Chemicals and Power		
F	Municipal Solid Waste to	Voung C C	John Wiley and Sons
5	Energy Conversion processes	Young G.C	John Wiley and Sons

14. SUGGESTED LEARNING WEBSITES

- a) <u>https://onlinecourses.nptel.ac.in/noc23_ch05/</u>
- b) <u>https://www.teriin.org/projects/green/pdf/National-Waste.pdf</u>
- c) <u>https://www.eai.in/ref/ae/wte/typ/clas/india_industrial_wastes.html</u>
- d) <u>Ministry of New & Renewable Energy Government of India (mnre.gov.in)</u>
- e) <u>http://www.ottusa.com/synthetic_fuel/synthetic_fuel</u>
- f) OSHA Technical Manual (OTM) | Occupational Safety and Health Administration
- g) <u>https://youtu.be/jYry2xe-HqY</u>
- h) <u>https://youtu.be/MuYdfxanAk8</u>
- i) <u>https://youtu.be/uXU2Pcxokb4</u>

15. PO-COMPETENCY-CO MAPPING

Semester V	Waste to energy conversion technology (4350502)							
Semester v	POs							
Competency & Course Outcomes	PO1 Basic & Discipline -specific knowledg e	PO2 Problem Analysis	PO3 Design/dev elopment of solutions	PO4 Engineerin g Tools, Experiment ation &Testing	PO5 Engineering practices for society, sustainability & environment	PO6 Project Manageme nt	PO7 Life- long learning	
<u>Competency</u>		To Superv	vise operation	n and to Optin	mize energy and	Other Invento	ries	
CO1: Identify different wastes as a energy source		3.00	1.00	-	1.00	1.00	3.00	
CO2: Apply concept of Thermo- chemical Conversion for biogas production	1.00	3.00	3.00	1.00	2.00	1.00	2.00	
CO3: Apply concept of Bio-chemical Conversion for biogas production	1.00	3.00	3.00	1.00	2.00	1.00	2.00	
CO4: Utilize algal biomass as source of energy production	1.00	2.00	2.00	-	2.00	1.00	2.00	

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE

GTU Resource Persons

Sr. No.	Name and Designation	Institute	Contact No.	Email ID
	Mr. SHAH PARTH SATISHBHAI	GOVERNMENT		parthgcet@gmail.com
1	(Lecturer in chemical	POLYTECHNIC,		
	Engineering)	GANDHINAGAR		
	Mr. SOSA VIDURKUMAR	GOVERNMENT		vidur.sosa@gmail.com
	PUNJABHAI	POLYTECHNIC,		
2	(Lecturer in chemical	RAJKOT		
	Engineering)			

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2023(COGC-2021) Semester-V

Course Title: Chemical Engineering Project -I

(Course Code: 4350503)

Diploma Programme in which this course is offered	Semester in which offered
Chemical Engineering	5 th Semester

1. RATIONALE

Project work serves as a means for students to utilize their coursework knowledge and skills to solve particular problems or execute projects, ultimately fostering innovative skills. In addition, Developing a plant for a chemical product is a complex task that requires a comprehensive report encompassing various aspects such as the chemical process and unit operations, properties of raw materials and products, economic factors, safety and pollution issues, and material and energy consumption. Chemical engineering students need to prepare such reports to become successful entrepreneurs while keeping in mind sustainability factors. A wide range of sustainable chemical products can be chosen from different sectors, including petrochemicals, fertilizers, pharmaceuticals, pesticides, natural products, polymers, and dyes. Careful consideration must be given to major equipment specifications, plant layout, and location to ensure the sustainability and success of the project. The syllabus provided is a guide, and instructors have the option to motivate students to develop prototypes, conduct experiments, or generate novel ideas that spark innovation.

2. COMPETENCY

The course should be taught and curriculum should be implemented with the aim to develop required skills so that students are able to acquire following competency:

• Identifying a chemical product, evaluating its essential features and characteristics, understanding the manufacturing processes involved, and selecting appropriate Process, equipment and instruments for the production.

3. COURSEOUTCOMES(COs)

The theory, practical experiences and relevant soft skills associated with this course are to be taught and implemented, so that the student demonstrates the following industry oriented COs associated with the above mentioned competency:

- a) Select a chemical product based on market demand, raw material availability, and potential profitability.
- b) Analyze the market and historical trends of a chemical product and explain its essential features and characteristics.
- c) Evaluate the properties and applications of chemical products in different contexts.
- d) Identify critical steps involved in production process.
- e) Select major equipments and instruments for chemical manufacturing processes.

4. TEACHINGANDEXAMINATIONSCHEME

Teaching		Total		Exam	ninationSche	eme					
Sche	me(InHo	ours)	Credits(L+T+P)	Theory Marks		Theory Marks		Theory Marks Practical Marks		Marks	Total Marks
L	Т	Р	С	СА	ESE	СА	ESE				
0	0	4	2	0	0	50	50	100			

Legends: L-Lecture; **T** – Tutorial/Teacher Guided Theory Practice; **P** - Practical; **C** – Credit, **CA** -Continuous Assessment; **ESE**-End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

The following practical outcomes (PrOs) are the sub-components of the COs. Some of the **PrOs** marked '*' (in approx. Hrs column) are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

Sr. No.	Practical/Exercise (Course Outcomes in Psychomotor Domain according to NBA Terminology)	Unit No.	Approx. Hrs Required
	selecting one chemical product based on specific criteria such as demand, availability of raw materials, and potential profitability	1	8
	Evaluate the historical development of the selected chemical product, including major breakthroughs, challenges	2	4
	Compare the chemical and physical properties of the selected chemical product with other similar products in the market	3	4
	Identify the various applications of the selected chemical product in different industrial and commercial contexts	3	8
	Analyze the advantages and disadvantages of each manufacturing process for the selected chemical product to determine the most suitable sustainable process.	4	12
	Create a flow diagram of the manufacturing process for the selected chemical product, including inputs and outputs at each step.	4	8
	select and use appropriate equipment and instruments for the selected chemical process	5	12
		Total	56

<u>Note</u>

- i. More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.
- ii. The following are some sample 'Process' and '#Product' related skills (more may be added/deleted depending on the course) that occur in the above listed Practical Exercises of this course required which are embedded in the COs and ultimately the competency.

Sr.No.	Sample Performance Indicators for the PrOs (The instructor is	Weight
	permitted to make slight modifications as deemed necessary).	age in %

1	Clarity and organization of the report	20
2	Demonstration of technical knowledge	25
3	Significance of problem/solution/conceptual feasibility analysis	15
4	Rigor and appropriateness of the methodology	15
5	Accuracy and relevance of the results	15
6	Analysis and interpretation of the results	05
7	Overall contribution to the field	05
	Total	100

6. MAJOREQUIPMENT/INSTRUMENTSANDSOFTWAREREQUIRED: N/A

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs and PrOs. More could be added to fulfill the development of this competency.

- a) Work as a leader/a team member.
- b) Follow ethical practices
- c) Observe safety measures
- d) Good house keeping
- e) Time management
- f) Practice environmentally friendly methods and processes.

The ADOs are best developed through laboratory/field-based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year
- ii. 'Organization Level' in 2nd year.
- iii. 'Characterization Level' in 3rd year.

8. COURSE DETAILS

Unit	Major Learning Outcomes(Course Outcomes in Cognitive Domain according to NBA terminology)	Topics and Sub-topics
Unit-I	1.1 Identify a chemical product	1. Choose a chemical product
Chemical Product Selection and	by conducting a market survey	from a range of chemical
Market Analysis		sectors, including but not
	1.2 selecting one based on	limited to petrochemicals,
	specific criteria such as demand,	fertilizers, pharmaceuticals,
	availability of raw materials,	pesticides, natural products,
	and potential profitability.	polymers, acids and alkalis,
		specialty chemicals, dyes, and pigments.

•

Unit-II Product survey and Industry Analysis	 2.1 Explain the essential features and characteristics of the selected chemical product. 2.2 Examine the current trends in the market and industrial landscape of the selected chemical product. 2.3 Evaluate the historical development of the selected chemical product, including major breakthroughs, challenges. 	 2.1 Introduction of the selected chemical product, including its essential features and applications 2.2 Historical development of the selected chemical product, including major milestones and breakthroughs. 2.3 The current state of the selected chemical product, including market trends and major industries involved in its production.
Unit-III Characteristics and Application of chemicals involved in process	 3.1 Discuss the chemical and physical properties of the raw materials used in the production of the selected chemical product, and their impact on the final product. 3.2 Compare the chemical and physical properties of the selected chemical product with other similar products in the market (if any). 3.3 Identify the various applications of the selected chemical product in different industrial and commercial contexts. 	 3.1 The properties of raw materials used in the production of the chemical product. 3.2 chemical and physical properties of the final product 3.3 Application of final product in different industrial and commercial contexts (Instructors are requested to stimulate students to think creatively and unconventionally).
Unit-IV Process Analysis for Sustainable Manufacturing	 4.1 Understand the different manufacturing processes involved in producing a chemical product and their respective steps, inputs, and outputs. 4.2 Analyze the advantages and disadvantages of each manufacturing process for the selected chemical product to determine the most suitable sustainable process. 4.3 identify the critical steps in the production process. 4.4 Create a flow diagram of the manufacturing process for the selected chemical product, including inputs and outputs at each step. 	 4.1 Different manufacturing processes with flow diagram involved in producing the chemical product and their respective steps, inputs, and outputs. 4.2 Merits, demerits, and engineering challengesof each manufacturing process for the selected product. 4.3 Various critical steps in the production process. 4.4 waste generation/environmental impact of each process.

Unit-V Selection of Major Process Equipments and Instrumentation	5.1 Identification of the necessary equipment and instruments for the selected chemical process	5.1 Description of the necessary equipment and instruments for the selected chemical process.5.2 Advantages and
	5.2 Understanding the functions and roles of each equipment and instrument in the manufacturing process	disadvantages of selected instruments. 5.3 The functions and roles of each equipment and instrument in the manufacturing process.
	5.3 select and use appropriate equipment and instruments for the selected chemical process	5.4 Selection criteria for instrument

9. SUGGESTEDSPECIFICATIONTABLEFORQUESTIONPAPERDESIGN:N/A

Legends: R=Remember, U=Understand, A=Apply and above (Revised Bloom's taxonomy)

<u>Note</u>: This specification table provides general guidelines to assist students for their learningand to teachers to teach and question paper designers/setters to formulate test items/questionsto assess the attainment of the UOs. The actual distribution of marks at different taxonomylevels(of R, Uand A) in thequestion paper mayslightlyvaryfromabovetable.

10. SUGGESTEDSTUDENTACTIVITIES

Following are the suggested student-related activities which can be undertaken to accelerate the attainment of the various outcomes in this course:

- 1. Industrial visit
- 2. Laboratories experiment
- 3. Literature Study
- 4. Attend Workshops
- 5. Internships
- 6.Take Part in Competitions
- 7.Course/topic based presentation

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Faculty can act as mentors to their students, providing guidance and support as they work on their final year projects.
- b) Faculty can facilitate group work, encourage peer feedback and provide

opportunities for students to work together on projects.

- c) Faculty can provide opportunities for students to work with laboratory equipment and conduct experiments, for example.
- d) Guide students to address issues on environment and sustainability with reference to using the knowledge of this course.
- e) Provide regular feedback and assessment on student work.
- f) Faculty can provide resources and support for students to pursue their own interests and areas of study.
- g) Faculty can provide opportunities for students to share their work with their peers and receive feedback.
- h) Faculty can incorporate active learning strategies, such as group discussions and problem-solving activities, into their final year project instruction.

12. SUGGESTEDMICRO-PROJECTS: N/A

13. SUGGESTEDLEARNINGRESOURCES

Sr. No.	TitleofBooks	Author	Publication with place, year and ISBN
1	Encyclopedia of Chemical Processing and Design	Jhon J. McKetta, William A. Cunninghalm	Marcel Dekker Inc., New York and Basel
2	Encyclopedia of Chemical Technology	Kirk and Othmer	John Wiley and Sons, Wiley Interscience
3	Ullman's Encyclopedia of Industrial Chemistry	Ullman	VCH Publishers, Germany
4	Chemical Process Technology Encyclopedia	Coincidine	McGraw-Hill
5	Perry's Chemical Engineers' Handbook	Robbert H. Perry, Down W. Green	McGraw-Hill
6	Plant Design and Economics for Chemical Engineers	Max Peters, Klaus Timmerhaus	McGraw Hill
7	Chemical Engineering Plant Design	Frank C. Vilbrandt, Charles E. Dryden	McGraw Hill
8	Chemical Engineering Design: Principles, Practice and Economics of Plant and Process Design	Gavin Towler, R. K. Sinnott	Butterworth-Heinemann
9	Process Engineering	James R. Couper	Marcel & Dekker Economics
10	Stoichiometry	B. I. Bhatt, S.M. Vora	Tata McGraw Hill
11	Safety and Accident Prevention in Chemical Operation	Faweett, Wood	Interscience Publishers
12	A course in Industrial Safety	K.U. Mistry	N.K.M. Publication

Pollution Control in Process Industries	S.P. Mahajan	Tata-McGrawHill
Safe Handling of Hazardous Chemicals	A.K. Rohatgi	J.K. Enterprise

14. SUGGESTEDLEARNINGWEBSITES

- a) <u>https://archive.nptel.ac.in/course.html</u>
- b) <u>https://chemicalengineeringworld.com</u>
- c)https://www.chemengonline.com/
- d)<u>https://chemicalengineeringsite.in/</u>

15. PO-COMPETENCY-CO MAPPING

Semester V		Ch	emical Engi	neering pro	ject –I (435050	03)	
Semester v				POs			
Competency &CourseOutc omes	PO1 Basic&Disc ipline- specifickno wledge	PO2 ProblemA nalysis	PO3 Design/deve lopmentofso lutions	PO4 EngineeringT ools, Experimenta tion &Testing	PO5 Engineeringprac tices forsociety,sustai nability& environment	PO6 ProjectManag ement	PO7 Life- longlearn ing
<u>Competency</u>	Superv	ise opera	ation and m	aintenance	of various hea	at transfer e	quipments
CO1: Select a chemical product based on market demand, raw material availability, and potential profitability.	3	1	-	1	1	-	1
CO2: Analyze the market and historical trends of a chemical product and explain its essential features and characteristics	1	2	1	2	-	-	3
CO3: Evaluate the properties and applications of chemical products in different contexts	2	-	-	-	2	-	2
CO4: Identify critical steps involved in production process	3	2	1	-	-	1	2
CO5:Select major equipment and instruments for chemical manufacturing processes.	2	-	2	1	1	1	1

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE <u>GTU Resource Persons</u>

Sr. No. Name and Designation Institute Contact No. Em	
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1	Mr. TARAKKUMAR CHANDRAKANTBHAI PADHIYAR (Lecturer in chemical Engineering)	SHRI K. J. POLYTECHNIC, BHARUCH	9879294415	tcp.gpv@gmail.com
2	Mr. JENISH DAMJIBHAI RUPAPARA (Lecturer in chemical Engineering)	GOVERNMENT POLYTECHNIC, GANDHINAGAR	7600694681	jenish.rupapara@gmail.com

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021(COGC-2021) Semester-V

Course Title: Summer Internship-II

(Course Code: 4350504)

Diploma Programme in which this course is offered	Semester in which offered
Chemical Engineering	5 th Semester

1. RATIONALE

Chemical Process Industries covert raw materials in to useful products using various processes and operations. The role of chemical engineer is to operate and handle various processes, operations and equipment effectively and ensuring safe work conditions and compliance health and safety regulation. Internship is educational and career development opportunities, providing practical experience in a field of chemical engineering. It will expose Technical students to the industrial environment, which cannot be simulated in the classroom and hence creating competent professionals for the industry. It may help students to acquaint themselves with the field they are interested in. Summer Internship-II offer students the chance to put what they are learning into action, in a real-world environment. It may help student to strengthen existing soft skills that are beneficial to career, in addition to scientific and technical skills like accountability, interpersonal skills, organization skills, problemsolving skills, teamwork, creative thinking and time management skills etc. Summer Internship-II is a great way to gain hands-on experience real-life engineering workplace and supplement their engineering education. This allows students to apply academic coursework, and to gain insight into the professional life of an engineer. It will help to optimize manufacturing process thereby achieving production target with an economical cost.

2. COMPETENCY

The course should be taught and curriculum should be implemented with the aim to develop required skills so that students are able to acquire following competency:

- Operate various unit operations and processes by using standard operating procedure in the chemical process industry.
- Get hands-on training, technical experience and opportunities to learn, understand and sharpen the technical / managerial skills in a real-world environment.

3. COURSE OUTCOMES(COs)

The practical experiences and relevant soft skills associated with this course are to be implemented, so that the student demonstrates the following industry oriented COs associated with the above mentioned competency:

- 1. Apply basic chemical engineering knowledge to understand working of various unit operations and processes in the chemical process industry.
- 2. Make use of standard operating procedures for various unit operations and processes in the chemical process industry.

- 3. Identify engineering problems and troubleshooting in the chemical process industry.
- 4. Identify the various safety measures and pollution control techniques used in the chemical process industry.

5. Develop lifelong learning in team management, communication skills and supplement technical knowledge.

]	Feaching	5	Total		Exan	nination Scl	neme	
Scher	ne(In H	ours)	Credits(L+T+P)	Theory	Marks	Practical	Marks	Total Marks
L	Т	Р	С	CA	ESE	CA	ESE	
0	0	0	3	00	00	50	50	100

4. TEACHING AND EXAMINATION SCHEME

Legends: L-Lecture; *T* – Tutorial/Teacher Guided Theory Practice; *P* - Practical; *C* – Credit, *CA* -Continuous Assessment; *ESE*-End Semester Examination.

5. MODE OF INTERNSHIP

• Internship in the industry

Internship in the industries/organization is to be arranged for the students of 5th semester based on internship guideline.

• Project based internship

Suitable topic related to Chemical Engineering like project based on industrial problem, working model of the equipment, Lab experiment and data analysis, and Design and analysis of the system/equipment, etc can be given as a mini project to the students of 5th semester.

6. INTERNSHIP GUIDELINES:

The summer internship-II for students in industries/organization is to be arranged in the beginning of 5^{th} semester/ as per schedule given by GTU.

- Time Duration: 6 Weeks
- Training area: Large/Medium/Small Scale Chemical Process/ Allied industries

The general suggestive procedure for arranging internship is given below:

Step 1: Request Letter/ Email from the office of Institute should send to industry to allot slots of 6 weeks during the schedule given by GTU as internship periods for the students. Students request letter/profile/ interest areas may be submitted to industries for their willingness for providing the internship.

Step 2: Industry will confirm the internship slots and the number of students allocated for internships via Confirmation Letter/ Email. In case the students arrange the training themselves, the confirmation letter will be submitted by the students in the department office of the institute.

Step 3: Students on joining internship at the concerned Industry / Organization submit the Joining confirmation Letters / Email to department office.

Step 4: Students will undergo summer internship-II at the concerned Industry / Organization. Faculty mentor(s) evaluate(s) the performance of students and Evaluation Report should be maintained in department office.

Step 5: Students will prepare and submit summer internship-II report after completion of internship.

Step 6: Summer Internship-II Certificate to be obtained from industry and submit it to department office.

7. ROLE AND RESPONSIBILITY OF STUDENT

Summer Internship-II is a student centric activity. Therefore, the major role is to be played by the student.

- The student should interact with mentor to suggest choice for training and suitable Industry/Organization. If students have any contact in the industry/Organization, then same may be utilizes for getting permission for summer Internship for themselves and their peers.
- The students have to fill the consent form duly signed by their parents and submitted to department office of the institute.
- The student should follow all rules/regulation and safety procedures of the Industry/Organization during internship period.
- It is the responsibility of student to collect information from industry about manufacturing process, working of equipment/Equipment specification, raw material, maintenance procedure, organization structure etc.
- In case they face any major problem in the industry such as an accident or any disciplinary issue, they should immediately report the same to the institute.
- The student should prepare a Summer Internship-II report along with Student's Diary/ Weekly Logbook and submit to department at the time of submission.

8. EXPECTED LEARNING AREA FOR THE STUDENT DURING SUMMER INTERNSHIP-II

The following areas are expected to learn by students during summer internship program.

- Organization hierarchy
- Manufacturing process of industry
- Reaction involved in the process.
- Flow diagram of the manufacturing process
- Major equipment involved in the process
- Major utilities used in the plant
- Instrumentation and Control system in the industry
- Raw materials and products handling/ MSDS
- Raw materials and products laboratory testing
- Standard operating procedure of the process/equipment
- Permit System followed in the industry
- Hazards associated in the Industry
- Safety measures followed in the industry
- Awareness about various PPEs used in the industry
- Pollution control and effluent treatment techniques followed in the industry

9. STUDENT'S DIARY/ WEEKLY LOGBOOK

• The main purpose of writing diary is to cultivate the habit of documenting and to encourage the students to search for details. It develops the students thought process and

reasoning abilities.

- The students should record observations, impressions, information gathered and suggestions given, if any. It should contain the sketches & drawings related to the observations made by the students.
- Student's Diary/ WEEKLY LOGBOOK should be submitted by the students along with internship report at the time of submission.
- For sample format of Student's Diary/ Weekly Logbook refer Annexure-3.

10. SUMMER INTERNSHIP REPORT

After completion of Internship, the student should prepare a comprehensive report to indicate what he has observed and learnt in the internship period. Student diary will also help to a great extent in writing the internship report.

The following is the suggestive format for the internship report.

- Cover Page
- Institute Certificate
- Industry completion Certificate
- Student Declaration
- Acknowledgement
- Content page
- Chapters
- Appendix (optional)
- References (optional)

Chapter-1: Introduction to industry/organization (Types of products and services, history, turnover, location and number of employee, Organization structure of industry/organization and general plant layout etc)

Chapter-2: Manufacturing process, reaction along with its process flow diagram

Chapter-3: Major equipments/instruments used in process industries with their standard operating procedure and troubleshooting

Chapter-4: Raw materials and products handling/MSDS

Chapter-5: Safety measures and Work permit system

Chapter-6: Plant Utilities, Instrumentation and Process control system

Chapter-7: Pollution control and effluent treatment techniques

Note: The chapters and content may be change based on Type of Industry/Organization and Area of Internship.

11. EVALUATION

It will be evaluated on the basis of the following criteria:

- Attendance and regularity in maintenance of the student's diary
- Adequacy, quality of information and purposeful write-up/presentation
- Organization of the information, format, drawings, sketches, other information recorded
- Variety and relevance of learning experience/ Depth of knowledge and skills
- Practical applications, relationships with basic theory and concepts taught in the course

12. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs. More could be added to fulfill the development of this competency.

- a) Work as a leader/a team member.
- b) Follow ethical practices
- c) Follow and Observe safety measures
- d) Good house keeping
- e) Time management
- f) Practice environmentally friendly methods and processes.

13. SUGGESTIVEFORMAT

The following are suggestive format for Summer Internship-II course.

Annexure-1: Student Registration From

Annexure-2: Request Letter from Institute to Industry/Organization

Annexure-3 Student's Diary/ Weekly Logbook

Annexure-4: Summer Internship-II Completion Certificate

Annexure-5: Student Declaration

Annexure-6: Summer Internship-II Report Format

Annexture-7: Suggested Evaluation Rubrics

14. PO-COMPETENCY-CO MAPPING

Semester V	Summer Internship-II (4350504) PO5						
Competency &Course Outcomes	PO1 Basic&Dis cipline- specifickno wledge	PO2 Problem Analysis	PO3 Design/ Develop ment Of solutions	PO4 Engineering Tools, Experimenta tion &Testing	PO5 Engineering practices for society,sustain ability& environment	PO6 Project Management	PO7 Life- long learni ng
CO1: Apply basic chemical engineering knowledge to understand working of various unit operations and processes in the chemical process industry	3.00	2.00	1.00	2.00	2.00	1.00	2.00
CO2: Make use of standard operating procedures for various unit operations and processes in the chemical process industry.	3.00	1.00	-	2.00	2.00	2.00	1.00

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CO3: Identify engineering problems and troubleshooting in the chemical process industry.	3.00	3.00	2.00	2.00	2.00	2.00	2.00
CO4: Identify the various safety measures and pollution control techniques used in the chemical process industry.	3.00	2.00	1.00	2.00	3.00	1.00	2.00
CO-5: Develop lifelong learning in team management, communication skills and supplement technical knowledge.		1.00	1.00	1.00	-	3.00	3.00

15. COURSE CURRICULUM DEVELOPMENT COMMITTEE

GTU Resource Persons

Sr. No.	Name and Designation	Institute	Contact No.	Email ID
1	Mr. Jagdish Dattatrya Raut (Lecturer in Chemical Engineering)	Government Polytechnic, Valsad		jagdish.raut@gmail.com
2	Mr. Parth Dhruvkumar Prajapati (Lecturer in Chemical Engineering)	Government Polytechnic, Valsad		pdprajapati266@gmail.com

Annexure-1: STUDENT REGISTRATION FROM

[Name of the Institute]

[Name of the Department]

STUDENT REGISTRATION FROM

Title of Internship	
Type of Industry/Organization	
Enrollment Number	
Student Name	
Student Details	Contact No.:
	Communication Address:
	Parents Contact No.:
Name of Institute	
Mentor Details (Institute)	Name:
(institute)	Designation:
	Mobile No:
	Email Address:
Industry /Organization Details	Name:
	Address:
	Email:
	Contact No.
	Website:
Mentor Details (Industry)	Name:
(Industry)	Designation:
	Mobile No:
	Email:

Student Signature

Institute Mentor Signature

Annexure-2: REQUEST LETTER FROM INSTITUTE TO INDUSTRY/ ORGANIZATION

[Institute Letter Head]

Date: DD/MM/YYYY

To The General Manager (HR)

Subject: Request for 6 Weeks Summer Internship-II for student(s) of Diploma-Chemical Engineering

Dear Sir,

In Diploma-Chemical Engineering program, Gujarat Technological University has introduced Summer Internship-II for the students in the beginning of 5th semester.

In view of the above, I request your good self to allow our following _____student(s) for internship training in your esteemed organization. Kindly accord your permission and give one week time for student(s) to join training after confirmation.

Sr	Name of student	Enrollment No.	
No.			

A line of confirmation will be highly appreciated. You are requested to provide Summer Internship completion certificate to above student(s) after completion of internship period.

With Warm regards,

Authorized Signatory Designation (Name of the Institute)



Annexure-3 STUDENT'S DIARY/ WEEKLY LOGBOOK

Week:	1 / 2 / 3 /	/4/5/6	Date:	to			
Name of Organization							
Department/Division							
Name of Plant Head/							
Supervisor/In-charge Main Topic: Process/Observation/Information/Drawing/Learning points							
Cian of Diam II 1/C							
Sign of Plant Head/Supervisor/In-charge:							

GTU - COGC-2021 Curriculum

Annexure-4: SUMMER INTERNSHIP-II COMPLETION CERTIFICATE

[Industry Letter Head]

Date: DD/MM/YYYY

TO WHOM IT MAY CONCERN

This is to certify that ______, Enrollment No.______, a student of chemical Engineering Department, <Institute Name>has successfully completed his/her summer internship-II in the field of <Internship Area/Department>from <Start Date>to<End Date> (Total number of Weeks: ____) under the guidance of <Industry Mentor/Plant Head>.

During the period of her/ his internship program with us, he / she had been exposed to different processes and was found sincere, enthusiastic and dedicated toward work assigned.

We wish him/her every success in his/her life and career.

For <Industry/Organization Name> Authorized Signature Designation



Annexure-5: STUDENT DECLARATION

[Name of the Institute] [Name of the Department]

DECLARATION

I/We hereby declare that the Summer Internship-II report submitted along with the Internship entitled **<Internship Title**>submitted in partial fulfillment for the degree of Diploma in Chemical Engineering, Gujarat Technological University, Ahmedabad, is a bonafide record of work carried out by me/us at<Industry/Organization Name> under the supervision of <Industry / Institute Mentor Name> and that no part of this report has been directly copied from any students' reports or taken from any other source, without providing due reference.

Name of Student

Signature

Annexure-6: SUMMER INTERNSHIP-II REPORT FORMAT

CoverPage<Removethis>

[SUMMER INTERNSHIP-II]

<FontSize18><1.5line spacing>

A REPORT ON <INTERNSHIP AREA/DEPARTMENT> <NAME OF INDUSTRY/ORGANIZATION>

<FontSize14>

Submitted by

<FontSize14><Italic>

[NAME OF THE CANDIDATE]

<FontSize16>

[GTU Enrolment Number]

<FontSize14>

In partial fulfillment for the award of the degree of

<FontSize14><1.5linespacing><Italic>

DIPLOMA IN CHEMICAL ENINEERING

<FontSize16>

in

[Name of the Department]

[Name of the Institute with City]

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Institute Logo

GTU Logo

Gujarat Technological University, Ahmedabad

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[Month,Year]

Following are the guidelines for the preparation Summer Internship-II Report.

PAPER: Use A4 (210mm X 297mm) Paper.

MARGINS: Margins for pages including the regular text should be as below:

Left : 1.25 Inches

Right : 1.0 Inch

Top : 1.0 Inch

Bottom: 1.0 Inch

PREPARATION OF CHAPTERS

Color:

All the text including Tables should be Black prints. However, Graphs and Figures can have color prints.

Font:

Chapter Headings : Times New Roman 16pts, bold prin	t all capitals
--	----------------

Section Headings : Times New Roman 14pts, bold print all capitals

Subsection Headings: Times New Roman 12pts, bold print leading capitals (only first letter in each word should be capital

Regular Text : TimesNewRoman12pts, normal prints

Special Text : Times New Roman Italics 12 pts (for foot notes, symbols, quotes, Mathematical notations...)

Annexture-7: SUGGESTED EVALUATION RUBRICS

Suggested Evaluation Rubrics (Institute/Industry Mentor)

Enrollment No: _____

Name of Student: _____

Date of Evaluation:

Sr.	Parameter	Excellent	Good	Average	Poor	Very Poor	Obtained Marks
No.	Mark	5	4	3	2	1	Warks
1	Student regularity,	5		5	2	1	
1	Proactiveness/						
	responsiveness towards						
	the given tasks during the						
	Internship period						
2	Regularity in maintenance						
2	of the diary/Log book						
3	Quality and organization of						
4	information						
4	Purposeful writeup,						
	content, drawing and						
	sketches in the internship						
~	report						
5	Depth of technical						
6	knowledge and skills Variety and relevance of						
0	learning experience						
7	Quality of presentation						
8	Development of soft skills						
0	like communication,						
	presentation, team						
	work/leadership and						
	management skills						
9	Application of basic theory						
	and concepts of chemical						
	engineering						
10	Quality of work/ Practical						
	applications in forms of Outcome achieved.						
		Marka Ob4	ain ad O-	4 of 50			
	Total Marks Obtained Out of 50 (Minimum Passing Marks: 20)						
	(Minimum Passing Marks: 20)						

Name of Mentor/Examiner:

Signature of Mentor/Examiner:

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021(COGC-2021) Semester-V

Course Title: Fertilizer Technology

(Course Code: 4350505)

Diploma Programme in which this course is offered	Semester in which offered
Chemical Engineering	5 th Semester

1. RATIONALE

Indian economy is dominated by agriculture sector. Synthetic fertilizers are must for producing good crops. Hence it is needed to provide comprehensive and balanced understanding of essential link between chemistry and the synthetic fertilizer industry. It is therefore vital for chemical engineers to understand for each fertilizer product, its flow diagram for Industry production. For this purpose chemical engineers should have skills for arranging treatment, reaction and separation steps in a flow diagram for variety of fertilizers including Nitrogenous fertilizers, Phosphatic fertilizer, Potash Fertilizer, Complex fertilizer and Bio fertilizers is essential. Hence this course is designed to achieve this objective.

2. COMPETENCY

The course content should be taught and curriculum should be implemented with the aim to develop required skills in the students so that they are able to acquire the following competencies:

• Supervise the different stages in fertilizer production

3. COURSEOUTCOMES(COs)

The theory should be taught and practical's should be carried out in such a manner that students are able to acquire different learning outcomes in cognitive, psychomotor and affective domain:

- i. Apply the concept of unit process and unit operations for manufacturing of various fertilizers
- ii. Characterize fertilizers on the basis of different properties.
- iii. Identify engineering problems in various fertilizers manufacturing.
- iv. Outline applications of various fertilizer.

4. TEACHING AND EXAMINATION SCHEME

٦	Teaching	S	Total		ExaminationScheme			
Sche	me(InHo	ours)	Credits(L+T+P)	Theory Marks		Theory Marks Practical Marks		Total Marks
L	Т	Р	С	CA	ESE	CA	ESE	
3	-	2	5	30*	70	25	25	150

(*):Out of 30marks under the theory CA,10marks are for assessment of the micro-project to facilitate the integration of COs, and the remaining 20 marks are the average of 2 tests to be taken during the semester for assessing the attainment of the cognitive domain UOs required for the attainment of the COs.

Legends: L-Lecture; T – Tutorial/Teacher Guided Theory Practice; P - Practical; C – Credit, CA -Continuous Assessment; ESE-End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

The following practical outcomes (PrOs) are the sub-components of the COs. Some of the **PrOs** marked '*' (in approx. Hrs column) are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

Sr. No.	Practical/Exercise (Course Outcomes in Psychomotor Domain according to NBA Terminology)	Unit No.	Approx. Hrs Required
	Prepare chart for fertilizer classification with chemical formula and nutrient content	I	02
	Estimate nutrient content (% N, %P2O, % K2O) in different fertilizers from their chemical formula	Ι	02
3	Estimate percentage of Nitrogen in Ammonium chloride by substitution method	Π	02
4	Estimate percentage of Nitrogen in Ammonium sulfate by substitution method	П	02
5	Estimate percentage of Nitrogen in Ammonium chloride by back titration	П	02
6	Estimate percentage of Nitrogen in Ammonium sulphate by back titration	П	02
7	Analysis of Urea by Formaldehyde method	П	02
	8 Estimate percentage of Nitrogen in Ammonium Chloride/Sulphate by Kjeldhal's method		02
9	Estimate biuret content in Urea sample by colour comparison	П	02
10	Estimate ratio from Ammonia to Phosphoric acid in DAP	III	02
11	Prepare potassium sulphate	IV	02
12	Prepare potassium chloride	IV	02
13	Prepare potassium nitrate	IV	02
14	Estimate percentage of Nitrogen in DAP by Formaldehyde method	V	02
15	Estimate percentage of Nitrogen in DAP by Kjeldhal's method	V	02
16	Preparation of Organic fertilizer	VI	02
		Total	28

Note:

- i. More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.
- ii. The following are some sample 'Process' and '#Product' related skills (more may be added/deleted depending on the course) that occur in the above listed Practical Exercises of this course required which are embedded in the COs and ultimately the competency.

Sr.No.	Sample Performance Indicators for the PrOs	Weight age in %
1	Handling of apparatus for precise measurements	10
2	Record observations correctly	20
3	Practice and adapt good and safe measuring techniques	10
4	Calculations, Interpretation of results and their conclusion.	20
5	Prepare report of practical in prescribed format	10
6	Solve assignment questions.	20

7	Viva-voce	10
	Total	100

6. MAJOR EQUIPMENT/INSTRUMENTS AND SOFTWARE REQUIRED

These major equipment/instruments and Software required to develop PrOs are given below with broad specifications to facilitate procurement of them by the administrators/management of the institutes. This will ensure the conduction of practice in all institutions across the state in a proper ways the desired skills are developed in students.

Sr.No.	Equipment Name with Broad Specifications	PrO. No.		
1	Hot Air Oven: Temperature is controlled by digital temperature indicator cum controller from ambient to 250°C with ± 0.1°C Accuracy. Power supply: 220/230V, 50Hz single phase, Capacity (Approx.): 50 – 100 liter, Type of Shelves: 03, Material of Inner Chambers: SS304, Material of Outer Chamber: MS with powder coated paint, Material of Shelves:SS wire mesh	All		
2	Laboratory Weighing Balance: Type of Laboratory Balance: Analytical, Sensitivity (mg): 1 mg, Maximum Capacity of weighing (grams): 200 g, Shape of PAN:Circular, PowerSupply: Single Phase, Display: LED.	All		
3	Hot Plate With Magnetic Stirrer: Number of stirring Positions:1,Calibration:Automatic Calibration, Magnetic stirrer with a hot plate, Speed Control Accuracy of sets peed (+/-) (RPM): 5, Maximum stirring capacity per position:3000ml,Top plate Material: Stainless steel	All		
4	Lab cooling bath: 220V/50HZ, 1.5KW, 370*340*480mm	All		
5	Kjeldahl apparatus: Flasks:30ml,50ml,100ml, Heating Element : Kanthal A-1, Heater Watt :200 watt ,Max. Temperature : 350°C	8,16		
6				
7	Accessories: Burner, Stand	All		
8	Chemicals: Ammonium chloride, Ammonium sulphate, Urea, DAP,KCL, KOH, NaCl, Indicators	All		

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of theabovementioned COs and PrOs. More could be added to fulfill the development of thiscompetency.

- a) Work as a leader/a team member.
- b) Follow ethical practices
- c) Observe safety measures
- d) Good house keeping
- e) Time management
- f) Practice environmentally friendly methods and processes.

The ADOs are best developed through laboratory/field-based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

i. 'Valuing Level' in 1st year

- ii. 'Organization Level' in 2nd year.
- iii. 'Characterization Level' in 3rd year.

8. UNDER PINNING THEORY

The major underpinning theory is given below based on the higher level UOs of *Revised Bloom'staxonomy* that are formulated for development of the Cos and competency. If required, more such higher-level UOs could be included by the course teacher to focus on the attainment of COs and competency.

Unit	Major Learning Outcomes(Course Outcomes in Cognitive Domain according to NBA terminology)	Topics and Sub-topics
Unit – I Overview of fertilizer	1a. Justify the need for synthetic fertilizer1b. Categorize fertilizers1c. Explain role of essential elements for plant growth	 1.1 Synthetic fertilizers 1.2 Classification of fertilizers 1.3 Role of essential Elements in plant Growth 1.3.1 Macro elements 1.3.2 Micro elements
	1d. Select the relevant fertilizers for the different types of crops	1.4 Application of fertilizers considering nutrient balance and types of crop
Unit – II Nitrogenous Fertilizers	 2a. Describe different properties of ammonia, nitric acid and urea 2b. Prepare synthesis path for manufacturing synthesis gas, ammonia, nitric acid and urea 2c. Describe ammonia synthesis converter 2d. Explain storage and transportation of ammonia 2e Estimate concentration of Nitric acid 2f. Describe the engineering problems of ammonia, nitric acid and urea manufacturing 2g. Describe the manufacturing process 	 2.1 Physical, chemical properties and applications of ammonia, nitric acid and urea 2.2 Manufacturing of ammonia synthesis gas 2.3 Explain sketch of ammonia synthesis converters 2.4 Manufacturing of ammonia 2.5 Storage and Transportation of Ammonia 2.6 Manufacturing of Nitric Acid 2.7 Concentration of Nitric acid by Mg(NO3) 2.8 Manufacturing of Urea 2.9 Major engineering problems of ammonia, nitric acid and urea manufacturing 2.10 Manufacturing of 2.10.1 Ammonium nitrate 2.10.2 Ammonium sulphate 2.10.3 Ammonium chloride
Unit – III Phosphatic Fertilizer	 3a. Describe various physical, chemical properties and uses of Phosphoric acid 3b. Describe the manufacturing process of Phosphoric acid by Wet process and Electric furnace method 3c. Describe the engineering problems of phosphoric acid 3d. Describe the manufacturing Superphosphate and Triple superphosphate 	 3.1 Physical, chemical properties and applications of phosphoric acid 3.2 Manufacturing phosphoric acid by Wet Process 3.2.1 Strong sulphuric Acid Leaching 3.2.2 Hydrochloric Acid Leaching 3.2.3 Electric Furnace Process 3.3. Major engineering problems of phosphoric acid 3.4. Manufacturing of Superphosphate

		3.5 Manufacturing of Triple superphosphate
Unit – IV Potassic Fertilizers	4a. Describe physical, chemical properties ,manufacturing and uses of potassium chloride, potassium nitrate and potassium sulphate	 4.1 Physical, chemical properties, manufacturing and uses of 4.1.1Potassium Chloride 4.1.2 Potassium nitrate 4.1.3 Potassium sulphate
Unit – V Complex Fertilizer	5a. Explain the manufacturing of NPK,ASP,CAN and DAP fertilizers with sketches	 5.1 Manufacturing of 5.1.1 NPK fertilizer 5.1.2 Ammonium Sulphate Phosphate (ASP), 5.1.3 Calcium Ammonium Nitrate(CAN) 5.1.4 Di-ammonium phosphate
Unit – VI Bio Fertilizer	 6a. Justify the need for biofertilizers and its benefits 6b. Describe the Nitrogen fixing and Phosphate solubilising biofertilizers 6c. Explain preparation a biofertilizers 6d.Explain organic fertilizers 	 6.1 Types of Biofertilizers 6.2 Biofertilizers: 6.2.1 Nitrogen-fixing biofertilizers 6.2.2 Phosphate solubilizing biofertilizers 6.3 Preparation of a biofertilizers 6.4 List out organic fertilizers 6.5 Outline advantages of organic fertilizers over conventional fertilizer

9. SUGGESTED SPECIFICATION TABLE FOR QUESTION PAPER DESIGN

	Unit Title	Taabiaa	Distribution of Theory Marks				
Unit		Teaching Hours	R	U	Α	Total	
		nours		Marks			
I	Overview of fertilizer	4	02	02	02	06	
II	Nitrogenous Fertilizers	14	06	10	06	22	
III	Phosphatic Fertilizer	8	04	06	04	14	
IV	Potassic Fertilizers	5	03	04	03	10	
V	Complex Fertilizer	6	02	06	02	10	
VI	Bio Fertilizer	5	02	04	02	8	
	Total	42	19	32	19	70	

Legends: R=Remember, U=Understand, A=Apply and above (Revised Bloom's taxonomy) <u>Note</u>: This specification table provides general guidelines to assist students for their learning and to teachers to teach and question paper designers/setters to formulate test items/questions to assess the attainment of the UOs. The actual distribution of marks at different taxonomy levels (of R, Uand A)in the question paper may slightly vary from above table.

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, following are the suggested student-related *co-curricular* activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform following activities in group and prepare reports of about 5 pages for each activity. They should also collect/record physical evidences for their (student's) portfolio which may be useful for their placement interviews:

Following is the list of proposed student activities like:

- 1. Assignments
- 2. Technical Quiz/MCQ Test
- 3. Presentation on some course topic

- 4. I-net based assignments
- 5. Undertake micro-Project in team/individually

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Massive open online courses (*MOOCs*) may be used to teach various topics/subtopics.
- b) Guide student(s) in undertaking micro-projects/activities.
- c) Different types of teaching methods i.e. video demonstration, activity based learning, case study, m-learning need to be employed by teachers to develop the outcomes.
- d) Some *of the topics/sub-topics* which are relatively simpler or descriptive are to be given to the students for *self-learning* but to be assessed using different assessment methods.
- e) Teachers need to ensure to create opportunities and provisions for *co-curricular activities*.
- f) Guide students to address issues on environment and sustainability with reference to using the knowledge of this course
- g) OERs, Vlab, and Olabs may be used to teach for the teaching of different concepts.

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her at the beginning of the semester. In the first four semesters, the micro-project is group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application-based, internet-based, workshop-based, laboratory-based, or field-based. Each micro-project should encompass two or more COs which are the integration of PrOs, UOs, and ADOs. Each student will have to maintain a dated work diary consisting of individual contributions in the project work and give a seminar presentation of it before submission. The duration of the microproject should be about **14-16** *(fourteen to sixteen) student engagement hours* during the course. The student sought to submit micro-project by the end of the semester (so that they develop industry-oriented COs.

A suggestive list of micro-projects is given here. This should relate highly to the competency of the course and the COs. Similar micro-projects could be added by the concerned course teacher.

- 1) Prepare a chart of the properties of given product
- 2) Prepare a chart to demonstrate manufacturing process.
- 3) Prepare a report on major engineering problem of given manufacturing process
- 4) Prepare a chart of application of given products
- 5) Prepare a power point presentation on a topic "List of Fertilizer manufacturing industries in India"
- 6) Prepare a PowerPoint presentation or animation showing different types of Fertilizer manufacturing Process
- 7) Library survey regarding fertilizers in different industries
- 8) Prepare a model of different fertilizers product flow diagram

13. SUGGESTEDLEARNINGRESOURCES

Sr. No.	TitleofBooks	Author	Publication with place, year and ISBN
1	Outlines of Chemical Technology, 3rd edition	M. Gopala Rao, Marshall Sitting	Affiliated East West Press (Pvt) Ltd-New Delhi
2	Shreve's Chemical Process Industries, 5th edition	Austin G.T.	McGraw Hill publication – New Delhi
	Chemical Technology – Vol. I , 2nd edition	G.N. Pandey and Shukla	Vani Books Company -Hyderabad
4	BioFertilizers and Organic Farming	Vayas S C and Modi H A	Akta prakashan, Nadiad

14. SUGGESTEDLEARNINGWEBSITES

i. http://nptel.ac.in/courses/103107086/4

ii.<u>http://ijset.com/ijset/publication/v1s6/285-291%20IJSET_PK%20JAGA.pdf</u>

iii.www.gses.com/images/pressreleases/Manufacturing-Process-Fertilizer.pdf

iv. http://nzic.org.nz/ChemProcesses/production/1A.pdf

v. http://tnau.ac.in/eagri/eagri50/SSAC222/lec12.pdf

vi. www.fnca.mext.go.jp/bf/bfm/pdf/Biofertilizer_Manual.pd

15. PO-COMPETENCY-CO MAPPING

SemesterIV	Fertilizer Technology (4350505)							
	POs							
Competency &Course Outcomes	PO1 Basic& Discipline- specific knowledge	PO2 Problem Analysis	PO3 Design/dev elopment of solutions	PO4 Engineering Tools, Experiment &Testing	PO5 Engineering practices for society, sustainability & environme nt	PO6 Project Manageme nt	PO7 Life-long learning	
Competency Supervise operation and main			nd mainten	ance of vario	ous equipn	nents		
CO1: Apply the concept of unit process and unit operations for manufacturing of various fertilizers	2.00	-	-	-	2.00	-	1.00	
CO2: Characterize fertilizers on the basis of different properties.	3.00	2.00	-	3.00	-	2.00	2.00	
CO3:. Identify engineering problems in various fertilizers manufacturing	2.00	1.00	-	1.00	1.00	-	-	
CO4: Outline applications of various fertilizer.	3.00	2.00	-	3.00	-	2.00	1.00	

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE GTU Resource Persons

Sr. No.	Name and Designation	Institute	Contact No.	Email ID
1	Mr. R.P. Hadiya (Lecturer in chemical Engineering)	G P Rajkot		rphadiya@yahoo.co.in
2	Mr. J D Rathod (Lecturer in chemical Engineering)	G P Valsad		Jdrathod94@gmail.com

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021 (COGC-2021) Semester-III

Course Title: Plant Utilities

(Course Code: 4330505)

Diploma programmer in which this course is offered	Semester in which offered
Plant Utilities	Third

1. RATIONALE

Diploma chemical engineer has to ensure smooth and proper operation of utilities and auxiliaries' plants such as steam, compressed air, instrumental air, inert gases, DM water and chilled water. These utilities are essential for manufacturing different chemical products. Use of concept of energy efficiency and green chemistry are necessary for energy conservation in chemical plant for producing materials of desired quality and to maintain plant safety. Hence the course has been design to develop these competencies and its associated cognitive and effective domain learning outcomes.

2. COMPETENCY

The course should be taught and implemented with the aim to develop required skills in students so that they are able to acquire following competency:

• Use different utilities in chemical process plants for various applications.

3. COURSE OUTCOMES (COs)

The practical exercises, the underpinning knowledge and the relevant soft skills associated with the identified competency are to be developed in the student for the achievement of the following COs:

- a) Select Various Methods For Water Softening And Purification.
- b) Explain Different Types of Steam Generators and Compressors along with their components.
- c) Select Refrigeration For Various Applications.
- d) Apply concepts of energy efficiency and green chemistry for conservation of utilities.

4. TEACHING AND EXAMINATION SCHEME

Теа	ching	Scheme	Total Credits	Examination Scheme				
	(In Ho	ours)	(L+T+P/2)	Theory	y Marks	Practica	Marks	Total
L	Т	Р	С	CA	ESE	СА	ESE	Marks
2	0	0	2	30	70	0	0	100

(*): For this practical only course, 50 marks under the practical CA have two components i.e. the assessment of micro-project, which will be done out of 10 marks and the remaining 15

marks are for the assessment of practical. This is designed to facilitate attainment of COs holistically, as there is no theory ESE.

Legends: L-Lecture; T – Tutorial/Teacher Guided Theory Practice; P -Practical; C – Credit, CA - Continuous Assessment; ESE -End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

Following practical outcomes (PrOs) are the sub-components of the Course Outcomes (Cos). Some of the **PrOs** marked **'*'** are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

Sr. No.	Practical Outcomes (PrOs)		Approx. Hrs. Required
	Not applicable		

<u>Note</u>

- *i.* More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.
- *ii. Care must be taken in assigning and assessing study report as it is a first year study report. Study report, data collection and analysis report must be assigned in a group. Teacher has to discuss about type of data (which and why) before group start their market survey.*

6. MAJOR EQUIPMENT/ INSTRUMENTS REQUIRED

These major equipments with broad specifications for the PrOs is a guide to procure them by the administrators to user in uniformity of practical's in all institutions across the state.

Sr.	Equipment Name with Broad Specifications	PrO. No.
No.		
1.	Not applicable	

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs and PrOs. More could be added to fulfill the development of this course competency.

- a) Follow ethical practices.
- b) Practice good housekeeping.
- c) Demonstrate working as a leader/a team member during brain storming.

The ADOs are best developed through the laboratory/field based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year
- ii. 'Organization Level' in 2nd year.

iii. 'Characterization Level' in 3rd year.

8. UNDERPINNING THEORY

The major Underpinning Theory is formulated as given below and only higher level UOs of *Revised Bloom's taxonomy* are mentioned for development of the COs and competency in the students by the teachers. (Higher level UOs automatically include lower level UOs in them). If required, more such higher level UOs could be included by the course teacher to focus on attainment of COs and competency.

Unit	Unit Outcomes (UOs)	Topics and Sub-topics
	(4 to 6 UOs at Application level)	
Unit – I	1.a Explain role of Utilities in	
Water as	Chemical Plant	1.1 List and use of various utilities in
Basic Utility	1.a.1 List various utilities in	chemical plant
	chemical plant & uses	1.2 Sources of water
	1.b List sources of Water	1.3 Hard & Soft water
	1.c Differentiate types of	1.4 Boiler Feed water and
	Water	demineralized water
	1.d Compare Softening	1.5 Methods of water softening
	processes of water	processes
	1.e Explain the process of	1.5.1 Lime soda process (Hot &
	Purification of water.	Cold)
	1.f Classify conventional and	1.5.2 Zeolite process
	green techniques for	1.5.3 Ion exchange process
	sterilization of water.	1.5.4 Phosphate process
		1.6 Purification of water
		1.6.1 Screening
		1.6.2 Sedimentation
		1.6.3 Coagulation
		1.6.4 Filtration
		1.6.5 Sterilization
		1.7 Conventional techniques
		1.7.1 Sterilization by chlorine
		1.7.2 Sterilization using
		bleaching powder
		1.7.3 Sterilization by
		chloramines solution
		1.8 Green techniques
		1.8.1 Sterilization using UV rays
		1.8.2 Sterilization using Ozon

Unit	Unit Outcomes (UOs)	Topics and Sub-topics
onic	(4 to 6 UOs at Application level)	
Unit – II	2.a Explain uses of utilities like	2.1 Use of Steam, Air & Inert Gases
Steam, Air &	Steam, Air & Inert Gases	as utilities
Inert Gases		
ment Gases	2.b Define properties of steam	2.2 Properties of steam 2.2.1 Enthalpy
	2.c Label the different part of	2.2.2 Wet steam
	steam generator	2.2.3 Saturated Steam
	2.d Classify steam generator	2.2.4 Superheated steam
	2.e Compare steam generators	2.2.5 Specific volume of steam
	2.f List the Factors affecting	2.3 Steam Generator : Classification,
	selection of Boiler	comparison , components, steam
	2.g Describe boiler accessories	handling, condensate removal
	and mountings for	2.4 Factors affecting selection of
	improving efficiency and	Boiler
	conservation of energy. 2.h Discuss utility air	2.5 Boiler Accessories and
	2.i Describe the working	mountings
	principle, application of Air	2.5.1 Air Pre heater
	compressors –	2.5.2 Super heater 2.5.3 Economizer
	2.i.1 Explain energy efficient	2.5.4 Steam trap
	alternative	2.6 Utility air
	2.j Describe properties of Inert	2.6.1 Compressed Air
	gases	2.6.2 Blower Air
	8	2.6.3 Fan Air
		2.6.4 Instrumental air
		2.7 Types of Air compressors
		2.7.1 Reciprocating Air
		compressors
		2.7.2 Rotary compressors 2.8 Energy efficient air compressor
		2.8 Energy enclent an compressor 2.8.1 Multistage compressors
		2.9 Inert gas - Nitrogen, Argon
Unit – III	3.a Explain the working principle	3.1 Concept of refrigeration
Refrigeration	of Refrigeration	3.2 Methods of Refrigeration
Reingerätion	3.b Distinguish methods of	3.2.1 Ice Refrigeration
	Refrigeration	3.2.2 Evaporative Refrigeration
	3.c Describe and TOR of	3.2.3 Vapor Refrigeration System
	refrigeration	3.3 TOR of refrigeration
	3.d Use primary and secondary	3.4 Types of Primary Refrigerants
	Refrigerants and list out	3.4.1 Ammonia
	green refrigerants	3.4.2 Halo Carbons (Freon of
	3.d.1 Explain advantages of green	Different type)
	refrigerants over	3.4.3 HFC (Hydro Fluorocarbon)
	conventional refrigerants	3.5 Types of secondary Refrigerants
		3.5.1 Water 3.5.2 Brine
		3.6 Selection of Refrigerants

Note: The UOs need to be formulated at the 'Application Level' and above of Revised Bloom's Taxonomy' to accelerate the attainment of the COs and the competency.

Unit	Unit		Distribution of Theory Marks			
	Title	Teaching Hours	R Level	U Level	A Level	Total Marks
I	Water as Basic Utility	10	10	10	6	26
	Steam, Air & Inert Gases	12	10	10	9	29
	Refrigeration	6	4	6	5	15
Total		28	24	26	20	70

9. SUGGESTED SPECIFICATION TABLE FOR QUESTIONPAPER DESIGN

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, following are the suggested studentrelated **co-curricular** activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform following activities in group and prepare reports of about 5 pages for each activity. They should also collect/record physical evidences for their (student's) portfolio which may be useful for their placement interviews:

- a) Undertake micro-projects in team/individually.
- b) Encourage Students for creating and designing water treatment processes using wastematerials.
- c) Students are encouraged to register themselves in various MOOCs such as: Swayam, edx, Coursera, Udemy etc to further enhance their learning.

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Guide student(s) in undertaking micro-projects.
- b) Diagnosing Essential Missed Learning concepts that will help for students to improve their performance.
- c) Guide Students to do Personalized learning so that students can understand the course material at his or her pace.
- d) Encourage students to do Group learning by sharing so that learning can be enhanced.
- e) About **20% of the topics/sub-topics** which are relatively simpler or descriptive in nature is to be given to the students for **self-learning**, but to be assessed using different assessment methods. Guide students on addressing the issues on environment and sustainability using the knowledge of this course.

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her in the beginning of the semester. In the first four semesters, the micro-project is group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application based, internet-based, workshop- based, laboratory-based or field-based. Each micro-project should encompass two or more COs which are in fact, an integration of PrOs, UOs and ADOs. Each student will have to maintain dated work diary consisting of individual contribution in the project work and give a seminar presentation of it before submission. The duration of the microproject should be about **14-16** *(fourteen to sixteen) student engagement hours* during the course. The students ought to submit micro-project by the end of the semester to develop the industry-oriented COs.

A suggestive list of micro-projects is given here. This has to match the competency and the COs. Similar micro-projects could be added by the concerned course teacher:

MICRO PROJECT 1: Identify sources of water at your college premises and measure thefollowing physical properties.

- 1. Measure temperature of water.
- 2. Measure TDS of water.
- 3. Measure pH of water.
- 4. Measure turbidity of water.

MIICRO PROJECT 2: Perform basic treatment techniques for purification of water.

MICRO PROJECT 3: Prepare 15-20 slides presentation showing classification of refrigeration & refrigerants.

13. SU	GGESTED LEARNING RESOURCES
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Sr. No.	Title of Book	Author	Publication with place, year and ISBN
1	Chemical Plant Utilities	Sathiyamoorthy- Manickkam	Lambert Academic Publishing;India, 2016, ISBN: 978-3-659-97828-9
2	Unit operation of Chemical Engineering	McCabe, Warren L., Julian C. Smith	McGraw Hill Publication; NewYork, 7th Edition, 2004
3	Plant utilities	D. B. Dhone	Nirali Prakashan; Pune, 2ndEdition, 2012
4	Power Plant Engineering	P.K. Nag	McGraw Hill Education; India,4th edition, 2017, ISBN: 978- 9339204044
5	Thermal Engineering	R.S. Khurmi, J. K. Gupta	S. Chand Publishing; India, 2008, ISBN: 9788121925730
6	Thermal Engineering	R.K.Rajput	Laxmi Publications; India, 11thedition,

Sr. No.	Title of Book	Author	Publication with place, year and ISBN
			2020, ISBN: 978- 8131808047

14. SOFTWARE/LEARNING WEBSITES

- a. https://nptel.ac.in/courses/112/107/112107291/
- b. https://www.thermodyneboilers.com/economisers/
- c. https://www.steamtrapefficiency.com/wp-content/uploads/BITHERM-STEAM-MANUAL.pdf
- d. http://www.silbert.org/MSA_WT_Manual.pdf
- e. http://ppuchem.blogspot.in/2013/02/unit-1-notes.html
- f. https://booksite.elsevier.com/samplechapters/9780080966595/Chapter_3.pd f

15. PO-COMPETENCY-CO MAPPING

Semester-III		Pla	ant Utilitio	-	ode: 4330505)		
Competency & Course Outcomes	PO 1 Basic & Discipline specific knowledge	PO 2 Problem Analysis	PO 3 Design/ develop- ment of solutions	POs PO 4 Engineering Tools, Experimen- tation &Testing	PO 5 Engineering practices for society, sustainability & environment	PO 6 Project Manage- ment	PO 7 Life-long learning
<u>Competency</u>	Use a	lifferent utili	ities in cher	nical process pl	ants for various	applicatio	ons.
Course Outcomes CO a) Select Various Methods For Water Softening And Purification.	3	2	2	1	2	1	2
CO b) Explain Different Types of Steam Generators and Compressors along with their Components.	2	1	-	-	1	1	1
CO c) Select Refrigeration For Various Applications.	3	1	2	-	2	2	3
CO d) Apply concepts of energy efficiency and green chemistry	3	2	2	1	3	2	3

Legend: '3' for high, '2' for medium, '1' for low and '-' for no correlation of each CO with PO.

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE

GTU Resource Persons

Sr. No.	Name and Designation	Institute	Contact No.	Email
1.	Mr. P M Gadhiya	GOVERNMENT POLYTECHNIC RAJKOT		gadhiyapiyush53@gmai.com
2.	Mr. I P Dave	GOVERNMENT POLYTECHNIC RAJKOT		ipd.fetr@gmail.com

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021 (COGC-2021) Semester-III

Course Title: Summer Internship-I

(Course Code: 4330001)

Diploma programme in which this course is offered	Semester in which offered	
All Branches of Diploma Engineering(Except Automobile, Bio	Third	
Medical, ICT, Power)	Tinid	

1. RATIONALE

Idea of Embedded Internships- AICTE has made 7-10 weeks summer internships mandatory in the new curriculum which will equip the students with practical understanding and training about industry practices in a suitable industry or organization. To make education holistic, sports, physical activities, values and ethics have been embedded in the curriculum.

We must agree that all Branches of Diploma Engineering are changing rapidly. New technologies are adding fast which effects can be seen in our society. Summer internship is a good option by which students to get flavor of such emerging technology and familiar with industry environment to identify scope and focus of their career development opportunities. Main objective of summer internship is hand-on practice to expose students for thinking about professional career by observing, understanding working mechanism of ongoing work of industry and to obtain various types of skills throughout internship program.

This two week mandatory internship is to equip the students with practical knowledge and provide them exposure to real time industrial environments. Further, in these internships, the option is provided to do internship in Government Agencies/ skill centers/ social sector/ Govt. initiated social schemes/ NGOs etc. The duration of internship will be two weeks. It will be after completion of 2nd Semester and before the commencement of Semester 3rd. Any options from following can be chosen by the students:

- 1. **Offline internship in industry** Student is supposed to produce joining letter for starting and relieving letter once the internship is over in case of Offline internship in any industry.
- 2. **Online internships** Student can select from any of approved /supported / recommended by the All India Council of Technical education for Internship (like Internshala/ NEAT/ Gujarat Knowledge Society Initiative etc.) or Approved by the state government or University approved
- 3. A Mini Project On some suitable topic related to respective branch. It can be small fabrication / experimental results/ simulations/ Application development / Design and / or Analysis of System(s) etc. depending on the branch of the student. Preferably a single student should carry out a mini-project.

2. COMPETENCY

- The purpose of this course is to help the student to attain flavor of the following industry identified competency through summer internship experiences:
- Develop multiple types of skills such as planning, communication, collaboration, decision making / Problem solving and management skills along with selected technical knowledge.

3. COURSE OUTCOMES (COs)

The practical exercises, the underpinning knowledge and the relevant soft skills associated with the identified competency are to be developed in the student for the achievement of the following COs:

- a) Learn and adopt the engineer's role and responsibilities with ethics.
- b) Get exposure to the industrial environment for professional activities.
- c) Get possible opportunities to learn, understand and sharpen the technical skills required for technical advancement.
- d) Develop managerial skills required for professional career.
- e) Attain skill for writing technical report and prepare poster for presentation.

4. TEACHING AND EXAMINATION SCHEME

Teach	ing Sc	heme	Total Credits Examination Scheme					
(Iı	1 Hour	:s)	(L+T+P/2)	Theory Marks		Marks Practical Mark		Total
L	Т	Р	С	CA	ESE	CA	ESE	Marks
0	0	0	1	0	0	25	25	50

- 1. **Offline internship in industry:** CA will be carried out based on submitted progress card by Industry resource person and ESE / Assessment will be carried out by institute resources person.
- 2. **Online internships:** CA will be carried out based on submitted certificate and ESE/ Assessment will be carried out by institute resources person.
- 3. A Mini Project: CA will be carried out based on project work by institute resources person.
- *Legends: L*-*Lecture; T Tutorial/Teacher Guided Theory Practice; P* -*Practical; C Credit, CA Continuous Assessment; ESE* -*End Semester Examination.*

List of Documents to be prepared for Submission:

- Detail report duly signed and approved by the internal/external mentor
- Presentation softcopy approved by the internal/external mentor
- Poster of summer internship activities approved by the internal/external mentor.

Sample forms for Registration and Evaluation of Summer Internship-I –SI-I are given below:

- 1) Both forms are mandatory to be filled at the commencement and completion of SI respectively.
- It is mandatory to file and map SI-I Registration and Evaluation with respective forms of SI-II (Later in Semester 5) so that students get enough exposure of industry / technology. (Mapping doesn't mean same industry/ company/ project-it can be independent/ different also.)
- 3) Mapping will be done to ease CA and ESE Evaluations.
- 4) A Seminar / Webinar can be arranged so that students coming from different industry / institute / project background can share experiences and learnings to their peers / all students of the same department.
- 5) Attached formats for Registration, Completion and Evaluation are suggestive. But, adhering to these formats is anticipated.

Summer Internship-I Registration Form

Note: Students needs to submit this registration form after finalizing mode of internship.

Student Details												
Enrollment Number												
Student Name												
6 1 5 1		T 1										
Student Details	Mobile I	Mobile Number :										
	Email A	ddress	:									
Branch												
Code of the Institute	Name o	of the	Instit	tute								
Mentor Details (Institute)	Name:											
	Designa	tion:										
	Mobile	No:										
	Email A	ddres	ss:									
Industry Details	Name:											
	Address	5:										
	Email:											
	Phone:											
	Website	e:										
Mentor Details (Industry)	Name:											
	Designa	tion:										
	Mobile	No:										
	Email A	ddres	SS									
Mode of Internship Carriec Out	l Online /	/ Offli	ne/ N	/lini P	roject							
Title of the Project/ Internship carried out												
Nature of Work Carried Out	Web De resul	-			on deve Analys	-), Exp	erimer	ntal
	Other pl	lease S	Speci	fy								

Student Signature

Faculty Signature

Summer Internship-I	-Suggested Letter for Completion
[Comp	any or Institute letter head]
No:	Date
TO WHOM	SO EVER IT MAY CONCERN
This is to certify that, Mr. /Mrs	
Enrollment No	Student of
Has successfully completed a two week	Internship in the field of
From the date:	to date:
[90% Attendance is n	nandatory for completion of Internship]
During the period of his/her summer interfollowing different processes and were 1	

Summer Internship-I -Evaluation Rubrics for Institute Evaluation Rubrics (Institute)

Enrollment No: _____

Branch:

Name of the Students:	
Date of Evaluation:	

Internal Evaluation – 25 Marks PA(I) (To be carried out by the mentor in consultation with Industry) Minimum Passing Marks: 13								
Parameter	Excellent	Good	Average	Not up the level of Satisfaction	Obtained			
Mark range	4-5	3-4	2-3	Below 2	Marks			
Knowledge acquisition in specific domain. 5 marks								
Skill and attitude attainment in specific domain. 5 marks								
Feedback and suggestions given are incorporated? 5 marks								
Quality of the prepared report and poster. 5 marks								
Quality of the presentation. 5 marks								
		То	tal Marks Obt	ained Out of 25 PA(I)				

Signature: ____

Institute Resource Examiner Name: ___

Suggested Evaluation Rubrics for Industry

Evaluation Rubrics (Industry)

Enrollment No: _____

Branch:

Name of the Students: ______ Date of Evaluation:

	External E	valuation – 2	5 Marks ES	E(V)	
(To be carried	out by the Ind	lustry Superv	visor) Minim	um Passing Marks: 1	13
Parameter	Excellent	Good	Average	Not up the level of Satisfaction	Obtained Marks
Mark range	4-5	3-4	2-3	Below 2	
Student regularity during the Internship period and proactive ness/responsiveness towards the given tasks (5 Marks) Work Plan, Execution and quality of work in forms of					
Outcome achieved (5 Marks)Engineering Tools and Techniques (5 Marks)					
Quality of poster design and presentation (5 Marks)					
Quality of the report and Skill (5 Marks)					
		Total 1	Marks Obtain	ed Out of 25 ESE(V)	

Signature: ___

Industry resource/ Examiner Name:

Common Note:

- 1) For Summer Internship / Projects / Seminar etc. Evaluation is based on work done, quality of report, performance in viva-voce, presentation etc. The internal / external assessment is based on the student's performance in viva-voce /work record respectively.
- 2) In case Industry Supervisor is not available / Institute Mentor/ Faculty can fill up both.

5. AFFECTIVE DOMAIN OUTCOMES

The following affective Domain Outcomes (ADOs) are embedded in many of the above mentioned COs. More could be added to fulfill the development of this course competency.

- a) Work as a leader/a team member as role of Engineer.
- b) Practice environmentally friendly methods and processes.

Follow safety precautions and ethical practices.

6. SUGGESTED STUDENT ACTIVITIES

- Following are the suggested student-related curricular, **co-curricular** activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform following activities and prepare reports and give presentation in front of students and faculty members. They should also collect/record physical evidences for their (student's) portfolio which may be useful for their placement interviews:
- a) Perform various tasks given by industry resources person during offline internship.
- b) Perform various tasks given during online internship.
- c) Perform various task required to complete mini project work under guidance of faculty member.
- d) Summer Internship program Interns are required to give a presentation before review committee consisting of a group of academic staff members.
- e) The review committee gives feedback and suggests possible improvements in the work.
- f) At the end of the program all the Summer Internship program Interns make a poster presentation of the work carried out. The poster presentation is open to the public. It is also evaluated by faculty members.
- g) A completion certificate will be issued to all Summer Internship program Interns only after the completion of internship tenure.

7. SOFTWARE / LEARNING WEBSITES

An internship is a short term work program usually offered to students by companies and institutes who require staff for assistance at junior levels. Thus for the students undergoing internship a professional learning experience is provided to benefit them in their skills as well as career. It will brush existing skills and provide exposure to new skills. Generally it is provided ay entry level in the industry.

Here is a suggestive list for reference only.

- <u>https://www.internshala.com</u>
- <u>https://swayam.gov.in</u>
- <u>https://nptel.ac.in/</u>
- <u>https://neat.aicte-india.org/</u>
- <u>https://www.edx.org/</u>
- <u>https://www.coursera.org/</u>
- <u>https://www.udemy.com/</u>
- https://www.linkedIn.com
- https://www.stumags.com
- <u>https://www.letsintern.com</u>
- <u>https://www.internship.com</u>
- <u>https://www.glassdoor.com</u>

8. PO-COMPETENCY-CO MAPPING

Semester III	Summer Internship (Course Code:4330001)								
Semester III	POs								
Competency & Course Outcomes	PO 1 Basic & Disciplin e specific knowled ge	PO 2 Proble m Analy sis	PO 3 Design/ developme nt of solutions	PO 4 Engineerin g Tools, Experimen tation & Testing	PO 5 Engineering practices for society, sustainability & environment	PO 6 Project Manageme nt	PO 7 Life- long learni ng		
<u>Competency</u>	Us	e principl	es of basic elec	tronics to main And equipm	ntain various elect ent	tronics circuits	;		
CO1) Learn and adopt the engineer's role and responsibilities with ethics.	2	1	1	1	1	1	1		
CO2) Get exposure to the industrial environment for professional activities.	1	1	1	1	1	1	1		
CO3) Get possible opportunities to learn understand and sharpen the technical skills required for technical advancement.	2	1	2	2	1	1	1		
CO4) Develop managerial skills required for professional career.	1	1	2	1	1	1	1		
CO5) Attain skill for writing technical report and prepare poster for presentation.	1	1	-	1	1	1	1		

Legend: '3' for high, '2' for medium, '1' for low and '-' for no correlation of each CO with PO.

9. COURSE CURRICULUM DEVELOPMENT COMMITTEE

Sr. No.	Name and Designation	Institute	Contact No.	Email
1	Jiger P. Acharya	GP, Ahmedabad	9429462026	jigeracharya@gmail.com
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3	Umang D. Shah	GP, Ahmedabad	9427686364	umang.shah111gp@gmail
Dog D	agaunaa Dangang			

BoS Resource Persons

Sr. No.	Name and Designation	Institute	Contact No.	Email
1	Shri U. V. Buch- BoS Member and Subject in- charge (EC)	G P Ahmedabad	9825346992	uvbuch@gmail.com

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021 (COGC-2021) Semester-III

Course Title: Mechanical Operation

(Course Code: 4330501)

Diploma programmer in which this course is offered	Semester in which offered
Chemical Engineering	Third

1. RATIONALE

The operations of chemical plants require use of material handling and size reduction equipments, screens, agitator, mixers, centrifuges, cyclones, filters, storage & conveying of solids and other mechanical separation equipments. Therefore students must have information about the principles, construction, working and application of these equipments so that they can plan for their efficient use in plants. In this course the students would also learn simple calculations to judge the performance of these equipments.

2. COMPETENCY

The course content should be taught and implemented with the aim to develop different types of skills so that students are able to acquire following competencies:

• Plan and supervise operation of mechanical operation equipments.

3. COURSE OUTCOMES (COs)

The practical exercises, the underpinning knowledge and the relevant soft skills associated with the identified competency are to be developed in the student for the achievement of the following COs:

- a) Use fundamentals of mechanical operation.
- b) Apply concept of size reduction, separation, agitation-mixing, Storage and Conveying of Solid & Fluid.
- c) Operate size reduction equipment, separators, agitators, mixers and conveyors.
- d) Calculate properties of solid particles, power consumption, and resistance, crushing law constants, efficiency and material balance.

Teach	ing Sc	heme	Total Credits	Examination Scheme				
(Ir	n Hour	s)	(L+T+P/2)	Theory Marks Practical Marks		Theory Marks		Total
L	Т	Р	С	СА	ESE	CA	ESE	Marks
3	0	4	5	30	70	50	50	200

4. TEACHING AND EXAMINATION SCHEME

(*): Out of 30 marks under the theory CA, 10 marks are for assessment of the micro-project to facilitate the integration of COs, and the remaining 20 marks are the average of 2 tests to be taken during the semester for assessing the attainment of the cognitive domain UOs required for the attainment of the COs.

Legends: L-Lecture; T – Tutorial/Teacher Guided Theory Practice; P - Practical; C – Credit, CA - Continuous Assessment; ESE - End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

The practical/exercises should be properly designed and implemented with an attempt to develop different types of practical skills (**Course Outcomes in psychomotor and affective domain**) so that students are able to acquire the competencies (**Programme Outcomes**). Following is the list of practical exercises for guidance.

Note: Here only Course Outcomes in psychomotor domain are listed as practical/exercises. However, if these practical/exercises are completed appropriately, they would also lead to development of Programme Outcomes/Course Outcomes in affective domain as given in a common list at the beginning of curriculum document for this programme. Faculty should refer to that common list and should ensure that students also acquire those Programme Outcomes/Course Outcomes related to affective domain.

Sr. No.	Practical Outcomes (PrOs)	Unit No.	Option	Approx. Hrs. Required	
1.	Measure volume surface mean diameter, mass mean diameter, number of particles using a sieve shaker.	I		4	
2.	Perform differential and cumulative screen analysis	111		4	
3.	Test Kicks law for crushing in Jaw crusher.	П		4	
4.	Test Bond's law for crushing in a Roll crusher.	П		4	
5.	Test Rittinger's law for grinding in a Ball mill and measure critical speed.			4	
6.	Measure efficiency and cut diameter of Cyclone Separator.	IV		4	
7.	Determine rate of settling by Sedimentation.	IV		4	
8.	Measure rate of filtration, filter medium & cake resistance in Gravity filtration.	IV	Any three	4	
9.	Measure rate of filtration, filter medium & cake resistance in Vacuum filtration.	IV		4	
10.	Measure cake resistance, filter media resistance in Pressure filtration.	IV		4	
11.	Measure rate of filtration, cake resistance, filter media resistance in Centrifuge.	IV		4	
12.	Measure efficiency of separation in froth flotation cell.		Any	4	
13.	Measure efficiency of separation in Magnetic separator.		two	4	
14.	Measure efficiency of separation in Electrostatic separator.	111		4	

15.	Evaluate mixing index in mixer.	V	4
16.	Measure power consumption in baffled and unbaffled Agitation vessel.	V	4

<u>Note</u>

More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.

6. MAJOR EQUIPMENT/ INSTRUMENTS REQUIRED

These major equipments with broad specifications for the PrOs is a guide to procure them by the administrators to use in uniformity of practical's in all institutions across the state.

Sr.	Equipment Name with Broad Specifications	PrO. No.
No.		
1	Sieve shaker – Sieve dia – 100 mm to 200 mm, no of sieve – 6- 8 and Pan, Opening – as per standards (micro or coarse particle)	1,2
2	Jaw crusher – 10-50 kg/hr capacity, Suitable for operation on 415V, 50Hz, 3 Phase, AC supply with energy meter	3
3	Laboratory Roll crusher – 5-25 kg/hr capacity, Suitable for operation on 415V, 50Hz, 3 Phase, AC supply with energy meter	4
	Laboratory Ball mill - 5kg capacity, Suitable for operation on 415V, 50Hz, 3 Phase, AC supply with energy meter	5
4	Cyclone separators – Product Particle as per requirement, Suitable for operation on 220V, Power: 1000W with energy meter	6
5	Batch Sedimentation Set up (Lab Scale Model using Glassware)	7
6	Gravity filter (Lab scale model using Glassware)	8
7	Vacuum Filter (Lab scale model using Glassware and Vacuum pump)	9
8	Laboratory filter Press - Suitable for operation on 220V, Power: 1000W	10
9	Basket centrifuge - Suitable for operation on 415V, 50Hz, 3 Phase, AC supply	11
10	Froth flotation Cell, 5-15 kg/hr capacity, Suitable for operation on 220V, Power: 1000W with energy meter	12
11	Magnetic separator suitable for operation on 220V, Power: 1000W	13
12	Electrostatic separator.	14

Sr.	Equipment Name with Broad Specifications	PrO. No.
No.		
13	Double cone mixer - Suitable for operation on 220V, Power:	15
	1000W	
14	Ribbon Blender- Suitable for operation on 220V, Power:	15
	1000W	
15	Sigma Mixer- Suitable for operation on 220V, Power: 1000W	15
16	Agitation vessel setup – Suitable for operation on 220V,	16
	Power: 1000W with energy meter (with removable baffles)	

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs and PrOs. More could be added to fulfill the development of this course competency.

- a) Follow ethical practices.
- b) Practice good housekeeping.
- c) Demonstrate working as a leader/a team member during brainstorming.

The ADOs are best developed through the laboratory/field based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year
- ii. 'Organization Level' in 2nd year.
- iii. 'Characterization Level' in 3rd year.

8. UNDERPINNING THEORY

The major Underpinning Theory is formulated as given below and only higher level UOs of *Revised Bloom's taxonomy* are mentioned for development of the COs and competency in the students by the teachers. (Higher level UOs automatically include lower level UOs in them). If required, more such higher level UOs could be included by the course teacher to focus on attainment of COs and competency.

Unit	Unit Outcomes (UOs)	Topics and Sub-topics		
Unit – I	1a. Define Unit Operation and	1.1 Fundamentals of Uni		
Fundamental of	Unit Process	Operation and Unit		
Mechanical	1b. Differentiate Unit Operation	Process		
Operation	and Unit Process	1.2 Examples of Unit		
		Operation and Unit		
			Process	
	1c. Describe specific properties	1.3 Define & Calculate		
	of solids	 Particle density and Bulk 		
		density		

		 Sphericity Equivalent diameter Specific surface area Volume surface mean diameter Mass mean diameter Shape factor Number of particles in solid
	2a. Explain size reduction with applications2b. Select Size reduction	2.1 Principles of Size reduction and its application2.2 Factors for selection of size
	equipments	reduction equipment
	2c. Calculate Energy and power requirement in comminution	2.3 Energy and power requirement in comminution
Unit– II	2d. Explain Empirical laws of size reduction and work Index	 2.4 Laws of size reduction: (i) Rittinger's law (ii) Bond's law (iii) Kick's law 2.5 Calculate Power required for size reduction using empirical laws and calculation of work Index.
Size Reduction	2e. Explain Different Size reduction equipments	 2.6 Principle, construction, working and Application of Jaw crusher Gyratory crusher Roll Crusher Ball mill Hammer mill
	2f. Compute Angle of nip for Roll Crusher	2.7 Derive equation of angle of nip2.8 Calculation of angle of nip forRoll crusher
	2g. Calculate Critical speed of Ball mill	2.9 Derivation of equation of critical speed for Ball Mill2.10 Calculation of operating speed and critical speed for Ball Mill
	2h. Explain Open and Close circuit grinding	2.11 Difference between open circuit and close circuit grinding
Unit– III Solid- Solid separation	3a. Explain Screening3b. Compare types of screen analysis	 3.1 Basics of Ideal and Actual Screen 3.2 Types of Screen Analysis Cumulative analysis Differential analysis
	3c. Derive formula for	3.3 Capacity and effectiveness of

	effectiveness of screen	screen
	3d. Calculate capacity and effectiveness of screen3e. Explain different screening equipment	 Derivation of formula for overall effectiveness of screen Calculation of capacity and effectiveness of screen 3.4 Principle, Construction, Working & Application of Trommel, grizzlies,
		vibrating screen
	3b. Explain different Solid separation equipments	 3.5 Principle, Construction, Working & Application of Hydraulic Jig Double cone classifier Electrostatic precipitator Magnetic separator Froth flotation cell
	3c. Select solid separation equipment	3.6 Factors affecting selection of equipment for solid separation
	 4a. Describe filtration and Differentiate constant rate and constant pressure filtration 4b. Characterize filter media 4c. Explain filter aids and its application 	 4.1 Basics of filtration Constant Rate filtration Constant Pressure filtration 4.2 Filter media and its characteristics 4.3 Filter Aid & its application
Unit– IV Solid- fluid Separation	olid- fluid 4d.Explain cake resistance, filter	 4.4 Cake Resistance, Filter medium Resistance Constant Rate filtration Constant Pressure filtration 4.5 Principle, construction, working and application of Filter Press Rotary Drum Filter Leaf Filter Basket Centrifuge
	4f. Explain sedimentation 4g. Draw batch sedimentation curve	 4.6 Basic of Sedimentation Hindered settling Free settling 4.7 Batch Sedimentation Experiment Interphase height Vs Time curve for batch sedimentation
	4h. Explain Gravity Thickener	4.8 Principle, construction,

	 4i. Explain Terminal settling velocity 4j. Describe Cyclone Separator 4k. Calculate Cut diameter and efficiency of cyclone 	 working and application of Gravity Thickener 4.9 Terminal settling velocity 4.10 Principle, construction, working and application of Cyclone Separator 4.11 Cut diameter and Efficiency ofCyclone Separator
Unit– V Agitation and mixing	 5a. Describe agitation and mixing 5b. Classify impellers 5c. Compare various impellers 5d. Explain vortex formation and prevention 5e. Explain factors affecting agitation 5f. Calculate power consumption 5g. Describe Agitated vessel 	 5.1 Define agitation and mixing 5.2 Classification of Impellers and brief explanation 5.3 Vortex formation and swirling 5.4 Methods for Prevention of Vortex 5.5 Factors affecting on agitation and Mixing 5.6 Calculation of Mixing index & Power Consumption 5.7 Principle, construction, working and application of Agitated vessel
	5h. Explain different Types of Mixers	 5.8 Principle, construction, working and application of Ribbon blender Kneaders Banbury mixer Muller mixer
Unit VI Storage & Conveying of Solid and Fluid	6a. Describe different Storage of Solids	6.1 Angle of repose6.2 Bulk storage6.3 Storage in bins and silos
	6b Explain different types of Conveyor	 6.4 Types of Conveyors Screw conveyors Belt conveyors Bucket elevators Pneumatic conveyor Hydraulic Conveyor

Note: The UOs need to be formulated at the 'Application Level' and above of Revised Bloom's Taxonomy' to accelerate the attainment of the COs and the competency.

Unit	Unit Title		Distribution of Theory Marks				
		Teachin	R	U	Α	Total	
		g	Level	Level	Level	Marks	
		Hours					
I	Fundamental of Mechanical		2	_	2	07	
	Operation	04	2	2	3	07	
II	Size Reduction	10	4	5	8	17	
- 111	Solid-Solid Separation	07	3	4	4	11	
IV	Solid-Fluid Separation	10	4	5	8	17	
V	Agitation and Mixing	07	3	5	3	11	
VI	Storage and Conveying	04	2	3	2	07	
	Total	42	18	24	28	70	

9. SUGGESTED SPECIFICATION TABLE FOR QUESTIONPAPER DESIGN

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, following are the suggested studentrelated **co-curricular** activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform following activities in group and prepare reports of about 5 pages for each activity. They should also collect/record physical evidences for their (student's) portfolio which may be useful for their placement interviews:

Following is the list of proposed student activities like:

- 1. Assignments
- 2. Technical Quiz/MCQ Test
- 3. Presentation on some course topic
- 4. I-net based assignments
- 5. Undertake micro-Project in team/individually
- 6. Students are encouraged to register themselves in various MOOCS such as: Swayam,
- edx, Coursera, Udemy etc to further enhance their learning

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Guide student(s) in undertaking micro-projects.
- b) Diagnosing Essential Missed Learning concepts that will help for students to improve their performance.
- c) Guide Students to do Personalized learning so that students can understand the course material at his or her pace.
- d) Encourage students to do Group learning by sharing so that learning can be enhanced.
- e) About **20% of the topics/sub-topics** which are relatively simpler or descriptive in nature is to be given to the students for **self-learning**, but to be assessed using different assessment methods. Guide students on addressing the issues on environment and sustainability using the knowledge of this course.

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her in the beginning of the semester. In the first four semesters, the micro-project is group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application based, internet-based, workshopbased, laboratory-based or field-based. Each micro-project should encompass two or more COs which are in fact, an integration of PrOs, UOs and ADOs. Each student will have to maintain dated work diary consisting of individual contributions in the project work and give a seminar presentation of it before submission. The duration of the micro project should be about **14-16** *(fourteen to sixteen) student engagement hours* during the course. The students ought to submit micro-project by the end of the semester to develop the industry-oriented COs.

A suggestive list of micro-projects is given here. This has to match the competency and the COs. Similar micro-projects could be added by the concerned course teacher:

- 1. Prepare a chart/model of Size reduction equipments along with their Principle and applications.
- 2. Prepare chart/model for various types of impellers used in Industry.
- 3. Interpret results of Cumulative Analysis and Differential Analysis using Microsoft excel.
- 4. Prepare chart for Vortex formation and swirling and Methods for their prevention.
- 5. Draw suitability chart for various factors for selection of various mechanical equipment.
- 6. Prepare 15-20 slides power point presentation showing classification, construction and working of any mechanical equipment.
- 7. Compare Screen Effectiveness by Manual & Mechanical Method using Literature/ experimental data by the use of Microsoft excel.
- 8. Prepare Laboratory set up for Batch Sedimentation.
- 9. Prepare Working model of any mechanical operation equipment.
- 10. Prepare a demonstrative model of any mechanical operation equipment.

13. 3	SUGGES	ARNING	RESOL	JRCES	

Sr. No.	Title of Book	Author	Publication with place, year and ISBN
1	Unit Operations of Chemical Engineering	McCabe and Smith	McGraw Hill Publications, New Delhi

Sr. No.	Title of Book	Author	Publication with place, year and ISBN
2	Introduction to Chemical Engineering	Badger W. L. and Banchero J. T	McGraw Hill Publications, New Delhi
3	Unit Operation –I	Gavhane K. A.	Nirali Prakashan, Pune
4	Mechanical Operations	Swain A.K., G.K.Roy	Tata McGraw Hill Publications, New Delhi
5	'Chemical Engineering' Vol II,	J.M. Coulson & J.F. Richardson 'Chemical Engineering' Vol II,	6th Ed. Elsevier, 2003
6	Transport Processes and Separation Process Principles'	C.G. Geankopolis	4th Ed, Prentice Hall India, 2003.

14. SOFTWARE/LEARNING WEBSITES

- a. https://ndl.iitkgp.ac.in
- b. https://onlinecourses.nptel.ac.in
- c. https://swayam.gov.in/explorer
- d. www.cheresources.com

15. PO-COMPETENCY-CO MAPPING

Semester-III	Mechanical Operation (Course Code:4330501)								
Competency & Course Outcomes	PO 1 Basic & Discipline specific knowledge	PO 2 Problem Analysis	PO 3 Design/ develop- ment of solutions	POs PO 4 Engineering Tools, Experimen- tation & Testing	PO 5 Engineering practices for society, sustainabilit y & environment	PO 6 Project Manage- ment	PO 7 Life-long learning		
<u>Competency</u> Select and Operate Mecha			Mechanical	Operation equi	ipment in the Cl	hemical inc	lustry.		
<u>Course Outcomes</u> CO a) Use fundamental of mechanical operation.	2	-	-	-	-	-	-		
CO b) Apply concept of size reduction, separation, agitation- mixing, Storage and Conveying of solid and Fluid.	2	2	2	2	2	2	2		

CO c) Operate size reduction equipment, separators, agitators, mixers and conveyors.	3	2	2	2	3	2	2
CO d)Calculate properties of solid particles, power consumption, and resistance, crushing law constants, efficiency and material balance	2	2	2	3	2	-	1

Legend: '**3'** for high, '**2**' for medium, '**1**' for low and '-' for no correlation of each CO with PO.

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE

Sr. No	Name and Designation	Institute	Contact No.	Email
1.	Mrs. Bhoomi Luckvindersingh Guleria	Government Polytechnic Gandhinagar	8128349140	bhoomitrivedi87@gmail. com
2.	Mrs. Bhumika Bhavin Makwana	Shri K.J. Polytechnic, Bharuch	9662032947	bhmkrn@gmail.com

GTU Resource Persons

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021 (COGC-2021) Semester-III

Course Title: Fluid Flow Operation

(Code: 4330502)

Diploma Programme in which this course is offered	Semester in which offered	
Chemical Engineering	3 rd Semester	

1. RATIONALE

Knowledge of fluid mechanics is essential for the chemical engineer because the majority of chemical-processing operations are conducted either partly or totally in the fluid phase. This course deals with basic concepts and principles in hydrostatics and hydrodynamics and their application in solving fluid - mechanics problems. Using various theoretical and practical concepts of fluid mechanics, power requirement for pumps, blowers and compressors can be determined and friction losses through pipes and fittings can also be calculated.

2. COMPETENCY

The aim of this course is to help the student to attain the following industry identified competency through various teaching learning experiences:

• To maintain flow of different fluids in the chemical plants according to the process requirement.

3. COURSE OUTCOMES (COs)

The theory, practical experiences and relevant soft skills associated with this course are to be taught and implemented, so that the student demonstrates the following industry oriented COs associated with the above mentioned competency:

- 1. Calculate the pressure difference using fundamental concept of fluid statics and carry out dimensional analysis.
- 2. Solve various fluid flow problems using governing equations.
- 3. Calculate Friction losses from changes in velocity or direction
- 4. Understand the concept of fluidization.
- 5. Select the metering equipments and fluid moving machinery for appropriate chemical engineering operations.

4. TEACHING AND EXAMINATION SCHEME

Tea	ching Sc	heme	Total Credits	Total Credits Examination Scheme		Examination Scheme		
	(In Hour	s)	(L+T+P)	Theory	Theory Marks Practical Marks		Total Marks	
L	Т	Р	С	СА	ESE	СА	ESE	
3	-	4	7	30*	70	50	50	200

(*): Out of 30 marks under the theory CA, 10 marks are for assessment of the microprojectto facilitate the integration of COs, and the remaining 20 marks are the average of 2 tests to be taken during the semester for assessing the attainment of the cognitive domain UOs required for the attainment of the COs.

Legends: L-Lecture; T – Tutorial/Teacher Guided Theory Practice; P - Practical; C – Credit, CA -Continuous Assessment; ESE - End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

The following practical outcomes (PrOs) are the sub-components of the COs. Some of the **PrOs** marked '*' (in approx. Hrs column) are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

Sr.	Practical/Exercise (Course Outcomes in Psychomotor	Unit	Approx.
No.	Domain according to NBA Terminology)	No.	Hrs Required
1.	Estimate pressure drop using U-tube manometer	Ι	4
2.	Identify types of flow by using Reynolds's apparatus	П	4
3.	Use Bernoulli's apparatus for mechanical energy balance		4
4.	Estimate viscosity of water using Hagen-Poiseuille's equation		4
5.	Estimate friction losses through pipe, fittings and valves	IV	4
6.	Estimate friction losses through packed bed	IV	4
7.	Conduct a performance test on reciprocating pump and plot the operating characteristics	V	4
8.	Conduct the performance test on centrifugal pump and plot the operating characteristics	V	4
9.	Estimate friction losses due to sudden expansion and contraction in flow area	IV	4
10.	Determine the Co-efficient of discharge for venturi meter	VI	4
11.	Determine the Co-efficient of discharge for orifice meter	VI	4
12.	Determine the Co-efficient of discharge of the Rotameter and obtain the calibration error	VI	4
13.	Determine the Co-efficient of discharge for notches	VI	4
14.	Measure minimum fluidization velocity through fluidized bed	VII	4
		Total	56

Note

i. More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.

ii. The following are some **sample** 'Process' and '#Product' related skills (more may be added/deleted depending on the course) that occur in the above listed **Practical Exercises** of this course required which are embedded in the COs and ultimately the competency.

Sr.No.	Sample Performance Indicators for the PrOs	Weightage in %		
1	1 Handling of apparatus for precise measurements			
2	Record observations correctly	20		
3	3 Practice and adapt good and safe measuring techniques			
4	Calculations, Interpretation of results and their conclusion.	20		
5	Prepare report of practical in prescribed format	10		
6	Solve assignment questions.	20		
7	Viva-voce	10		
	Total	100		

6. MAJOR EQUIPMENT/ INSTRUMENTS AND SOFTWARE REQUIRED

These major equipment/instruments and Software required to develop PrOs are given below with broad specifications to facilitate procurement of them by the administrators/management of the institutes. This will ensure the conduction of practice in all institutions across the state in a proper way so that the desired skills are developed instudents.

Sr.No.	Equipment Name with Broad Specifications	PrO. No.
1	U-tube manometer:- Metering Tube : Special Uniform bore thick walled Borosilicate Glass Tube, End Connection :Nozzle Type, Mounting : wall panel mounting, Fluid : Mercury, Range : 250-0-250 mm, Accuracy : + 0.5 % of FSD	01
2	Reynolds's apparatus: - Test Pipe: Acrylic, 25 mm dia., 800 mm long, Test pipe attached with water tank having capacity of 200 liter, Dye Reservoir : 500 ml	02
3	Bernoulli's apparatus: - Testing Duct: MOC-Acrylic , Type : Circular with divergent and convergent sections, Inlet Tank and Outlet Tank: Capacity : 20 lit., MOC : SS – 304 With provision for overflow and valve to stabilize the flow. Pump: MOC : SS body Type : Monoblock Power : 0.5 HP, Single phase Delivery Size : 1/2"	03
4	Friction loss apparatus: - Header pipe connected with four pipes, First pipe MS 0.5 inch provided with ball valve, gate valve, globe valve, union joint, Second pipe 0.5 inch without fittings for viscosity determination, Third pipe 02 inch and 0.5 inch connected by sudden expansion and contraction, Forth pipe 01 inch provided with rotameter(Range:3–30 LPH), All pipes and fittings provided with pressure tapping at suitable places. Manometer: Type : U tube, Length : 500 mm, Fluid : Mercury filled up to 200 mm in each limb.	4,5,9,12
5	Packed bed apparatus:- Packed Column : Material Borosilicate Glass with both end made of Stainless Steel Dia. 48 mm approx., Height 750 mm approx, Packing Material: glass spherical particles	06

	Size 8-10mm approx, Water tank : Material Stainless Steel, Capacity 30 Ltrs, Water Flow Measurement : By Rotameter, Pressure Drop Measurement: By Manometer, Electricity supply: 1 Phase, 220 V, 50 Hz AC, 5-15 amp combined socket with earth connection.	
6	Centrifugal pump test rig: - Pump: Centrifugal Type, 2800 RPM, Drive: Variable speed, Sump Tank : Made of Stainless Steel, Compatible capacity, Measuring Tank : Made of Stainless Steel, Compatible capacity, Stop Watch : Electronic Pressure Gauge : Bourdon type, Range: 0 - 4 kg/ cm ² , Vacuum Gauge : Bourdon type, Range: 0-760 mm of Hg Compound Gauge : Bourdon type, Range: -760 mm of Hg to 2 kg/cm ² , RPM measurement: RPM Indicator with Proximity sensor, Electricity supply: Single Phase, 220 VAC, 50 Hz, 5-15 Amp combined socket with earth connection	07
7	Reciprocating pump test rig:- Pressure Gauge:Bourdon type, RPM measurement: Digital RPM Indicator with Proximity sensor., Tanks Material:Stainless Steel, Tank Capacity:40 Liters., Sump Tank Material:Stainless Steel, Sump Tank Capacity:65 Ltrs., Pump Capacity:1 HP, Electricity Supply: Single Phase, 220 V AC, 50 Hz, 5-15amp, Head:5 kg/cm2, Type of Pump : Double Acting cylinder of RPM 320,Piston Stroke : 4cm, Piston Diameter : 4.5cm, Suction pipe : 1", Delivery pipe : 3/4",Pressure gauge : 2 kg/cm2, Vacuum gauge : 0-760mm Hg.	08
8	Venturi and Orifice meter Apparatus:- Sump Tank: Capacity : 80 liter MOC : SS – 304 With ½" drain valve, Collection Tank: Capacity : 50 liter, MOC : SS - 304, With self graduated glass tube level indicator & 1" drain valve,Pump: Type : Monoblock MOC : SS – 304 Power : 1 HP, Single Phase,Venturimeter: Size : Suitable for 1" pipeline Dia. Ratio : 0.6- 0.64 MOC : Brass With suitable Pressure tapings Arrangement, Orificemeter: Size : Suitable for 1" pipeline Dia. Ratio : 0.6-0.64 MOC : SS 304 With suitable Pressure tapings, Rotameter: Size : Suitable for 1" pipeline Range : 3 – 30 LPH Metering tube: Borosilicate Glass Valve : Needle Valve provided integral, Manometer: MOC : Borosilicate glass tube Size : 500 mm, Fluid : Mercury filled upto 200 mm in each limb, Necessary piping and valves shall be provided to supply water to various flow meters.	10,11
9	Discharge Over Notches Apparatus:- Flow Channel: MOC : Rectangular Open Channel of Acrylic, Size : 600 (L) * 250 (W) * 180 (H) mm, Notches: MOC : SS – 304 (Thickness – 1.6 mm), Mounting : Interchangeable, Type: 1.Rectangular, 2. 45°V, Measuring Tank: Capacity : 40 liter, MOC : SS – 304 with self graduated glass tube level indicator, Level gauge: Type : Hook and Screw gauge, Pump: MOC : SS, Type : Monoblock, Power : 0.5 HP, Single Phase	13
10	Fluidized bed apparatus:- Material Borosilicate Glass with both end made of Stainless Steel Dia. 48 mm (approx.), Height 750 mm	14

(approx.) • Packing : Glass Beads • Water tank : Material Stainless
Steel, Capacity 30 Ltrs. • Water Circulation : FHP Pump,
Crompton/Sharp make. • Water Flow Measurement : By Rotameter
Pressure Drop Measurement : Manometer, Electricity supply: 1
Phase, 220 V AC, 50 Hz, 5-15 amp combined socket with earth connection.

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs and PrOs. More could be added to fulfill the development of this competency.

- a) Work as a leader/a team member.
- b) Follow ethical practices
- c) Observe safety measures
- d) Good house keeping
- e) Time management
- f) Practice environmentally friendly methods and processes.

The ADOs are best developed through laboratory/field-based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year
- ii. 'Organization Level' in 2nd year.
- iii. 'Characterization Level' in 3rd year.

8. UNDERPINNING THEORY

The major underpinning theory is given below based on the higher level UOs of *Revised Bloom's taxonomy* that are formulated for development of the COs and competency. If required, more such higher-level UOs could be included by the course teacher to focus on the attainment of COs and competency.

Unit	Major Learning Outcomes (Course Outcomes in Cognitive Domainaccording to NBA terminology)	Topics and Sub-topics
Unit – I Fluid Staticsand its Applications	 1a. Define Ideal fluid andReal fluid 1b. Differentiate between fluid statics and dynamics 	1.1 Ideal fluid and Real fluid1.2 Fundamentals of fluid statics and dynamics
	1c. Properties of fluid	 1.3 Define pressure, Static pressure, Dynamic pressure,Vacuum, Density, Viscosity (Absolute and Kinematic), Specific gravity
	1d. Compare compressible and incompressible fluids	1.4 Compressible and incompressiblefluids

	1e. Derive and apply equation	1.5 Derivation of equation of pressure in
	of pressure in static fluid.	static fluid
	1f. Describe different types of	1.6 Principle construction and workingof:
	manometers	Simple U-tube manometer, Inclined
	1g. Derive equation ofpressure	manometer
	difference	1.7 Derive equation of pressure
	1h. Explain differential pressure	difference for U-tube manometer
	cell	1.8 Differential pressure cell
	1.i Solve simple numerical	1.9 Numerical based on pressure and
		manometer
Unit– II	2a. Explain velocity change	2.1 Velocity field, velocity gradient, shear
Fluid–Flow	across cross section	stress and rate of shear
Phenomena	2b. Explain effect of solid	2.2 Boundary layer & it's formation
and basic	boundary	In straight tubes
equations	2c. Compare Newtonian and	2.3 Newtonian and Non-Newtonian
of fluid flow	Non-Newtonian fluids with	fluid
	2d. Describe Reynold'sexperiment	2.4 Reynold's experiment and Reynolds Number
	2e. Solve simple numerical	2.5 Numerical based on Reynolds Number
Unit– III	3a. Define velocities	3.1 Average velocity and mass velocity
Basic	3b. Derive continuity	3.2 Continuity equation for mass
Equations	equation	balance in steady flow
of Fluid	3c. Derive Bernoulli's	3.3 Bernoulli's equation and correctionsin
Flow	equation	Bernoulli's equation
	Major Learning Outcomes	
	(Course	
	Outcomes in Cognitive	Topics and Sub-topics
Unit	_	
	Domainaccording to NBA	
	terminology)	
	3d. Use Hagen-Poiseuille's Equation	3.4 Hagen-Poiseuille's Equation
	3e. Solve simple numerical	3.5 Numerical based on basic equations
Unit–IV	4a. Describe roughness of	4.1 Roughness of pipe
Friction	pipe	
in Flowing	4b. Explain hydraulic radius and equivalent diameter	4.2 Hydraulic radius and equivalent diameter
Fluid	4c. Use friction factor chart	4.3 Friction factor chart
	4d. Explain friction losses	4.4 Friction losses from: (a) Sudden
	p	expansion of cross section (b) Sudden
		contraction of cross section
		4.4.1 Friction loss in fittings and valves
Unit– V	5a. Compare pipe and tube	5.1 Introduction of pipe and tube
Transportation		5.2 Birmingham wire gauge(BWG) and
of fluid		schedule no.

of fluid

schedule no.

5.3 Types and uses of fittings and joints

5b. Describe fittings & joints

i -							
	5c. Select valve for particular	5.4 Construction and working of valves like:					
	application	(a) Gate valve (b) Globe valve					
		(c) Check valve (d)Butterfly valve					
		(e) Ball valve (f) Control valve(Air to					
		open and Air to close)					
	5d. Classify pumps	5.5 Classification of pumps					
	5e. Explain pumps with their	5.6 Construction and working of					
	startup shut and down	centrifugal, reciprocating and rotary					
	procedure	pump					
	5f. Explain characteristics of	5.7 Explain characteristics of centrifugal					
	centrifugal pump	pump					
	5g. Calculate NPSH, head	5.8 Developed head and power					
	and power for	requirement in centrifugal pump					
	centrifugal pump	5.9 NPSH, and Cavitation in					
		centrifugal pump					
		5.10 Numerical based on NPSH,					
		5.10 Numerical based on NPSH, efficiency, head and power 5.11 Introduction to Fan, Blowers,					
	5h. Explain Fan, Blower,						
	Compressor, Vacuum	Compressors, Vacuum pump and					
	pump and jet ejectors	jet ejectors					
Unit– VI Flow	6a. Classify flow measuring	6.1 Classification of flow measuring					
Measurement	devices	devices					
weasurement							
	6b. Describe and select flow	6.2 Construction, working principles					
	meters	and application of flow meters like					
		Venturi meter, Orifice meter,					
		Rotameter, weirs, Magnetic flow					
		meter					
	6c. Derive equation of flow	6.3 Derivation of equation of flow rate					
	rate	through Orifice meter, Venturi					
		meter.					
	Major Learning Outcomes						
	(Course						
	•						
Unit	Outcomes in Cognitive	Topics and Sub-topics					
	Domainaccording to NBA						
	terminology)						
	6e. Solve simple numerical	6.4 Numerical of Orifice meter, Venturi meter					
Unit– VII Fluidization	7a Explain Fluidization	7.1 Fluidization and its industrial applications					
	7b Discuss Drag, Drag coefficient, Stokes' law	7.2 Drag and drag coefficient Stokes' law					
	7c Explain the condition of fluidization.	7.3 Condition for Fluidization					
	7d Define minimum fluidization velocity	7.4 Minimum fluidization velocity					

9. SUGGESTED SPECIFICATION TABLE FOR QUESTION PAPER DESIGN

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· GTU - COGC-2021 Curriculum

Unit	Unit Title	Teaching Hours	R Level	U Level	A Level	Total Marks
Ι	Fluid Statics and itsApplications	06	02	03	04	09
П	Fluid–Flow Phenomena	05	02	03	02	07
	Basic Equations of Fluid Flow	06	02	06	02	10
IV	Friction in Flowing Fluid	04	02	02	02	06
V	Transportation of Fluid	10	05	08	05	18
VI	Flow Measurement	08	04	07	04	15
VII	Fluidization	03	01	02	02	05
	Total	42	18	31	21	70

Legends: R=Remember, U=Understand, A=Apply and above (Revised Bloom's taxonomy)

Note: This specification table provides general guidelines to assist students for their learning and to teachers to teach and question paper designers/setters to formulate test items/questions to assess the attainment of the UOs. The actual distribution of marks at different taxonomy levels (of R, U and A) in the question paper may slightly vary from above table.

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, the following are the suggested student- related *co-curricular* activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform the following activities in group and prepare small reports of about 5 pages for each activity. They should also collect/record physical evidence such as photographs/videos of the activities for their (student's) portfolio which will be useful for their placement interviews:

- a) Make a survey on types of valves and pumps we use domestically.
- b) Prepare a report on fluid flow through open channel in your city/village.
- c) Draw and report water distribution system of chemical engineering department building.
- d) Make a chart/poster on Mechanical seal and stuffing box.
- e) Determine the energy consumption and efficiency of existing pumps of laboratory.
- f) Collect different parts of various pumps from scrap yard.

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Massive open online courses (*MOOCs*) may be used to teach various topics/subtopics.
- b) Guide student(s) in undertaking micro-projects/activities.
- c) Different types of teaching methods i.e. video demonstration, activity based learning,

case study, m-learning need to be employed by teachers to develop the outcomes.

- d) Some *of the topics/sub-topics* which are relatively simpler or descriptive are to be given to the students for *self-learning* but to be assessed using different assessment methods.
- e) Teachers need to ensure to create opportunities and provisions for *co-curricular activities*.
- f) Guide students to address issues on environment and sustainability with reference to using the knowledge of this course
- g) OERs, Vlab, and Olabs may be used to teach for the teaching of different concepts.

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her at the beginning of the semester. In the first four semesters, the micro-projectare group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application-based, internet-based, workshop- based, laboratory-based, or field-based. Each micro-project should encompass two or more COs which are the integration of PrOs, UOs, and ADOs. Each student will have to maintain a dated work diary consisting of individual contributions in the project work and give a seminar presentation of it before submission. The duration of the microproject should be about **14-16** *(fourteen to sixteen) student engagement hours* during the course. The students ought to submit micro-project by the end of the semester (so that they develop industry- oriented COs.

A suggestive list of micro-projects is given here. This should relate highly to the competency of the course and the COs. Similar micro-projects could be added by the concerned course teacher.

- a) Prepare a model to demonstrate flow through open channels.
- b) Prepare a chart on different types of vacuum pumps.
- c) Prepare a chart on various types of manometers.
- d) Prepare a power point presentation on classification and demonstration of valves.
- e) Develop a friction factor chart experimentally.
- f) Prepare a model/chart/poster on pneumatic and hydraulic conveying.
- g) Prepare a chart that classifies fluid on various bases.
- h) Prepare a power point presentation on blowers and compressors.
- i) Prepare a power point presentation on classification and demonstration of flow measuring devices.
- j) Prepare a power point presentation on various methods to join pipes and tubes.
- k) Make a report on Cavitation, its identification and prevention.
- I) Prepare a power point presentation pertaining to friction loss in fittings and valves.
- m) Make a chart on selection criteria of pumps and valves.

13. SUGGESTED LEARNING RESOURCES

Sr. No.	Title of Books	Author	Publication with place, year and ISBN
1	Unit Operations of ChemicalEngineering	McCabe, Warren L.,Julian C. Smith	McGraw Hill Publication, New York 2004 (Seventh Edition)
2	Introduction to ChemicalEngineering	L.Badger, Julius T. Banchero	McGraw Hill Publication, New York 2004 (Seventh Edition)
3	Unit Operations of Chemical Engineering Vol-I	Chattopadhyay, P.	Khanna Prakashan, New Delhi, 1996
4	A text book of Fluid Mechanics	Khurmi, R.S.	S. Chand Publication, New Delhi 2002
5	Unit Operation –I	Gavhane, K.A.	Nirali Prakashan, Pune 2009
6	A Textbook Of Fluid Mechanics And Hydraulic Machines	Dr. R. K. Bansal	Publisher: Laxmi Publications, 2005 ISBN 10: 8131808157 ISBN 13: 9788131808153
7	Fluid Mechanics: Fundamentals and Applications	John. M. Cimbala Yunus A. Cengel	McGraw Hill Publication, New York 2006 (1 st edition) ISBN 0–07–247236–7

14. SUGGESTED LEARNING WEBSITES

- a) https://ndl.iitkgp.ac.in/
- b) https://www.vlab.co.in/
- c) https://nptel.ac.in/
- d) http://www.nzifst.org.nz/unitoperations/flfltheory.htm
- e) https://www.slideshare.net/AjinkyaKhandizod/fluid-flow-operations-applications-of-fluid-mechanics-rheological-classifications-of-fluid

15. PO-COMPETENCY-CO MAPPING

Semester III	Fluid Flow Operation(4330502)						
POs							
Competency & Course Outcomes	PO 1 Basic & Discipline- specific knowledge	PO 2 Problem Analysis	PO 3 Design/ developmen t of solutions	PO 4 Engineering Tools, Experimen- tation &Testing	PO 5 Engineering practices for society, sustainability & environment	PO 6 Project Management	PO 7 Life-long learning

<u>Competency</u>	Use p	orinciples	of Fluid Flov	w Operation	in chemical en	gineering app	lications
CO1. Calculate the pressure difference using fundamental concept of fluid statics and carry out dimensional analysis.	3	3	-	3	-	2	2
CO2. Solve various fluid flow problems using governing equations.	3	3	-	3	-	-	2
CO3. Calculate Friction losses from changes in velocity or direction	2	2	-	2	-	1	-
CO4.Understand the concept of fluidization.	2	-	-	2	-	1	-
CO5. Select the metering equipments and fluid moving machinery for appropriate chemical engineering operations	3	3	2	3	3	3	2

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE

GTU Resource Persons

Sr. No.	Name and Designation	Institute	Contact No.	Email ID
1	Ms. Y S Patel Lecturer in Chemical Engg.	G P Gandhinagar		yaminipatel2016@gmail.com
2	Mr. J D Kanani Lecturer in Chemical Engg.	G P Rajkot		jatinpatel5005@gmail.com

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021 (COGC-2021) Semester-III

Course Title: Process Calculation

(Course Code: 4330503)

Diploma programmer in which this course is offered	Semester in which offered
Chemical Engineering	Third

1. RATIONALE

Process calculation provides the fundamental information to determine the material and energy balances for all types of unit operations and unit processes across the equipment and overall chemical plant. Material and energy balance calculations are of prime importance for design and also for conservation of mass and energy to reduce the losses and cost that enhances overall economy of plant. The unit conversions, material and energy balance are the essential part in the practice of other courses such as mechanical operations, fluid flow, heat Transfer, mass transfer etc. Thus this course is a core course for chemical engineers and should be learned sincerely by students.

2. COMPETENCY

The purpose of this course is to help the student to attain the following industry identified competency through various teaching learning experiences:

• Determine material and energy balance for different unit operations and processes

3. COURSE OUTCOMES (COs)

The practical exercises, the underpinning knowledge and the relevant soft skills associated with the identified competency are to be developed in the student for theachievement of the following COs:

- a) Apply stoichiometric principles for solving chemical engineering problems.
- b) Calculate material balance for chemical process.
- c) Calculate energy balance for chemical process.
- d) Estimate amount of fuel and amount of air required for combustion process.

Teaching Scheme		heme	Total Credits	Examination Scheme						
(Ir	n Hour	s)	(L+T+P/2)	Theory Marks		Theory Marks		Theory Marks Practical Marks		Total
L	Т	Р	С	СА	ESE	СА	ESE	Marks		
4	0	0	4	30	70	0	0	100		

4. TEACHING AND EXAMINATION SCHEME

(*): Out of 30 marks under the theory CA, 10 marks are for assessment of the micro-project to facilitate integration of COs and the remaining 20 marks is the average of 2 tests to be taken during the semester for the assessing the attainment of the cognitive domain UOs required for the attainment of the COs.

Legends: CI-Class Room Instructions; T – Tutorial/Teacher Guided Theory Practice; P -Practical; C – Credit, CA - Continuous Assessment; ESE - End Semester Examination.. Legends: L-Lecture; T – Tutorial/Teacher Guided Theory Practice; P -Practical; C – Credit, CA -Continuous Assessment; ESE - End Semester Examination.

5. SUGGESTED PRACTICAL EXERCISES

Following practical outcomes (PrOs) are thesub-components of the Course Outcomes (Cos). Some of the **PrOs** marked **'*'** are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

Sr. No.	Practical Outcomes (PrOs)	Unit No.	Approx. Hrs. Required
	Not Applicable		

Note

i.More **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The above table is only a suggestive list.

ii. The following are some **sample** 'Process' and 'Product' related skills (more may be added/deleted depending on the course) that occur in the above listed **Practical Exercises** of this course required which are embedded in the COs and ultimately the competency

The following are some **sample** 'Process' and 'Product' related skills (more may be added/deleted depending on the course) that occur in the above listed **Practical Exercises** of this course required which are embedded in the COs and ultimately the competency.

Sr. No.	Sample Performance Indicators for the PrOs	Weightage in %
1	Question answer or Writing steps exercise	30
2	Executing of exercise	30
3	Result	40
		100

6. MAJOR EQUIPMENT/ INSTRUMENTS REQUIRED

This major equipment with broad specifications for the PrOsis a guide to procure them by the administrators to user in uniformity of practical's in all institutions across the state.

Sr. No.	Equipment Name with Broad Specifications	PrO. No.
1.	Not Applicable	

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample*Affective Domain Outcomes (ADOs) are embedded in many of the above-mentioned COs and PrOs. More could be added to fulfill the development of this course competency.

a) Work as a leader/a team member.

- b) Follow ethical practices
- c) Practice environmentally friendly methods and processes (environmental related).

The ADOs are best developed through the laboratory/field based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year
- ii. 'Organization Level' in 2nd year.
- iii. 'Characterization Level' in 3rd year.

8. UNDERPINNING THEORY

The major underpinning theory is given below based on the higher level UOs of *Revised Bloom's taxonomy* that are formulated for development of the COs and competency. If required, more such UOs could be included by the course teacher to focus on attainment of COs and competency.

Unit	Unit Outcomes (UOs)	Topics and Sub-topics
	(4 to 6 UOs at different levels)	
Unit-I Unit Systems	 1a. Explain the importance of process calculation. 1b. Define different unit systems. 1c. Explain the importance of physical quantities of Units. 1d. Convert units among different systems. 	 1.1 Introduction to process calculation 1.2 Dimensions and systems of units 1.3 Fundamental quantities of units, Derived quantities 1.4 Definition and units of force, volume, pressure, work, energy, power, heat 1.5 Unit conversions in FPS, MKS and SI systems
Unit– II Basic Chemical Calculations	2a. Calculate important physical quantities.2b. Calculate composition of mixtures and solutions.	 2.1 Definition and calculations of mole, atomic weight, molecular weight, equivalent weight, specific gravity and API gravity 2.2 Expression of composition of mixtures and solutions 2.3 Morality, Normality, Morality, gm/lit and related simple numerical
Unit-III Ideal Gas Law	3a. Derive ideal gas law.3b. State reference conditions.3c. Calculate important quantities for ideal gas mixture.	 3.1 Concept of ideal gas 3.2 Derivation of ideal gas law 3.3 STP and NTP conditions 3.4 Dalton's law, Amagat's law, Raoult's Law and Henry's Law 3.5 Relation between mole%, volume% and pressure% of ideal gases 3.6 Calculation of average molecular weight, density, mole%, weight% in gas mixture in SI/MKS systems
Unit– IV	4a. Explain law of conservation of mass.	4.1 Law of conversation of mass4.2 Brief description and simple

Unit	Unit Outcomes (UOs)	Topics and Sub-topics
Onic	(4 to 6 UOs at different levels)	
	4b. Calculate mass balance of	material balance calculation of
Material	important unit operations at steady	drying, distillation, absorption,
	state condition.	mixing, crystallization, evaporation
Balance In	4c. Describe purging, recycling and	4.3 Single stage material balance
Processes	bypassing operations.	calculation of leaching and
Without		extraction
Chemical		4.4 Brief idea regarding recycling,
Reactions		purging and by passing operation
Unit– V	5a. Explain basic concepts of	5.1 Definition: Limiting reactant, Excess
	material balance with chemical	reactant, conversion, yield and
Material	reaction.	selectivity
Balance In	5b. Calculate mass balance with	5.2 Simple numerical for finding yield,
Processes	chemical reaction.	conversion and composition
Involving		5.3 Simple calculation of material
Chemical		Balance based on reaction.
Reactions		
Unit– VI	6a. Calculate heat capacity, specific	6.1 Heat capacity and specific heat
	heat, heat capacity of gas mixture	6.2 Mean heat capacity of gases
Energy Balance	and liquid mixture.	6.3 Heat capacity of gas mixture and
	6b. Explain concepts of sensible heat	liquid mixture
	and latent heat.	6.4 Calculations of heat capacity by
	6c. Calculate standard heat of	integral equation up to three terms
	formation and heat of reaction.	6.5 Brief explanation of sensible and
		latent heat of fusion, sublimation, vaporization
		6.6 Calculations of standard heat of
		formation from heat of combustion
		data
		6.7 Calculations for heat of reaction
		from heat of formation and heat of
		combustion data
		6.8 Hess's Law and calculations
Unit– VII	7a. Describe combustion.	7.1 Introduction of combustion
	7b. Describe calorific value.	7.2 Types of fuels
Combustion	7c. Calculate calorific value and air	7.3 Calorific values of fuels
	requirement for combustion.	7.4 Proximate and ultimate analysis of
		solid fuel
		7.5 Numerical related to calorific
		values of fuel from composition 7.6 Numerical related to air
		Requirement and composition of flue gases
		flue gases.

9. SUGGESTED SPECIFICATION TABLE FOR QUESTIONPAPER DESIGN

Unit	Unit Title	Teaching	Distrib	oution of	Theory	Marks
No.		Hours	R	U	Α	Total
			Level	Level	Level	Marks

1	Unit Systems	4	2	2	2	06
2	Basic Chemical Calculations	7	2	2	4	08
3	Ideal Gas Law	7	2	2	4	08
4	Material Balance In Processes Without Chemical Reactions	10	0	6	7	13
5	Material Balance In Processes Involving Chemical Reactions	8	2	3	7	12
6	Energy Balance	12	2	4	8	14
7	Combustion	8	2	2	5	09
ΤΟΤΑ	TAL 56 12 21 37		70			

Legends: R=Remember, U=Understand, A=Apply and above (Revised Bloom's taxonomy)

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, following are the suggested studentrelated **co-curricular** activities which can be undertaken to accelerate the attainment of the various outcomes in this course: Students should perform following activities in group and prepare reports of about 5 pages for each activity. They should also collect/record physical evidences for their (student's) portfolio which may be useful for their placement interviews:

- a) Group assignments based on mass and energy balance of equipments like heat exchanger, boilers, distillation column, evaporator, dryer, reactors, absorption column.
- b) Use of MS-Excel in solving numerical.
- c) Draw block diagram and write down overall and component material balance for various mass transfer operation and mechanical operations.

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Massive open online courses (*MOOCs*) may be used to teach various topics/sub topics.
- b) Guide student(s) in undertaking micro-projects.
- c) *'L' in section No.* 4 means different types of teaching methods that are to be employed by teachers to develop the outcomes.
- d) About **20% of the topics/sub-topics** which are relatively simpler or descriptive in nature is to be given to the students for **self-learning**, but to be assessed using different assessment methods.
- e) With respect to *section No.10*, teachers need to ensure to create opportunities and provisions for *co-curricular activities*.
- f) Guide students for reading data sheets.

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her in the beginning of the semester. In the first four semesters, the micro-projects

are group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application based, internet-based, workshop-based, laboratory-based or field-based. Each micro-project should encompass two or more COs which are in fact, an integration of PrOs, UOs and ADOs. Each student will have to maintain dated work diary consisting of individual contribution in the project work and give a seminar presentation of it before submission. The duration of the micro project should be about **14**-**16** *(fourteen to sixteen) student engagement hours* during the course. The students ought to submit micro-project by the end of the semester to develop the industry-oriented COs.

A suggestive list of micro-projects is given here. This has to match the competency and the COs. Similar micro-projects could be added by the concerned course teacher:

- a) Give a data of different unit operation and calculate material balance.
- b) Give a data of different unit operation and calculate energy balance.
- c) Prepare chart on molecular weight and equivalent weight.
- d) Visit of chemical process plant: Prepare block diagram showing material balance for process equipment used in plant which you have visited.

Sr. No.	Title of Book	Author	Publication with place, year and ISBN
1	Stoichiometry	B I Bhatt and S B Thakore	McGraw Hill Education; 5th edition (1 July 2017), ISBN: 978-0070681149
2	Basic Principles and Calculations in Chemical Engineering	Himmelablau David M.	PHI Learning, New Delhi, Year-2003, ISBN: 9789332549623
3	Stoichiometry and Process Calculations	Narayanan K.V. and Lakshmikutty B	PHI; 2nd edition,Year- 2016ISBN: 8120352890
4	Introduction to Process Calculations (Stoichiometry)	K. A. Gavhane	NiraliPrakasan, Pune, 2015

13. SUGGESTED LEARNING RESOURCES

14. SOFTWARE/LEARNING WEBSITES

- https://nptel.ac.in/courses/103103165
- Basic Principles & Calculations in Chemical Engg (CD Rom)
- https://www.unitoperation.com/

15. PO-COMPETENCY-CO MAPPING

Semester III	Process Calculation (Course Code : 4330503)
	POs

Competency & Course Outcomes	PO 1 Basic & Discipline specific knowledge	PO 2 Problem Analysis	ment of	PO 4 Engineering Tools, Experimenta tion&Testing	PO 5 Engineering practices for society, sustainability & environment	PO 6 Project Manage ment	PO 7 Life-long learning
<u>Competency</u>	Test variou				components and do g safe work practic	-	relevant
Course Outcomes CO 1) Apply stoichiometric principles for solving chemical engineering problems.	3	1	2	1	1	-	-
CO 2) Calculate material balance for chemical process.	3	3	3	2	1	-	-
CO 3) Calculate energy balance for chemical process.	3	2	3	2	1	-	1
CO 4) Estimate amount of fuel and amount of air required for combustion process.	3	2	3	2	1	_	1

Legend: '3' for high, '2' for medium, '1' for low and '-' for no correlation of each CO with PO.

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE

GTU Resource Persons

S. No.	Name and Designation	Institute	Contact No.	Email
1.				

NITTTR Resource Persons

S. No.	Name and Designation	Department	Contact No.	Email
1.	Mr. Harsh B Shukla, Lecturer in K.J.Polytechnic, Bharuch	Chemical Engineering		shuklahb22@gmail.com
2.	Mr. Chetan Panchal, Lecturer in G.P. Gandhinagar	Chemical Engineering		chetanpanchal91@gmail.com

GUJARAT TECHNOLOGICAL UNIVERSITY(GTU)

Competency-focused Outcome-based Green Curriculum-2021 (COGC-2021) Semester-III

CourseTitle: Chemical Process Technology (Course Code: 4330504)

Diploma programme in which this course is offered	Semester in which offered
Chemical Engineering	Third

1. RATIONALE

The importance of this subject arises from the need of providing comprehensive and balanced understanding of essential link between chemistry and the chemical industry. It is vital to develop simple but meaningful flow diagram for each chemical product which a student can understand. This course develops skill for arranging and understanding treatment, reaction and separation steps in a flow diagram for variety of chemicals including acids, chloro-alkalis, fuels and industrial gases, cement, lime, polymer, dyes and intermediates, pharmaceutical, fermentation, pesticides, Soap and detergents, fertilizerand many other products. Diploma holders utilize this skill to read and recognize each steps of process flow diagrams during their job. The area of job may be production, R and D, design, technical services, project development, sales and marketing etc.

2. COMPETENCY

The course content should be taught and implemented with the aim to develop different types of skills leading to the achievement of the following competencies:

• Synthesize reactions and unit operations steps to develop and operate a chemical plant to manufacture important chemicals.

3. COURSEOUTCOMES(COs)

The practical exercises, the underpinning knowledge, and the relevant soft skills associated with this competency are to be developed in the student to display the following COs:

At the end of the course, a student will be able to

- 1) Explain the classification and properties of various chemicals
- 2) Apply concept of the manufacturing processes of various chemicals with neat sketch to operate chemical plant.
- 3) Identify major engineering problems encountered in manufacturing processes.
- 4) Suggest applications of various chemicals.

4. TEACHING AND EXAMINATION SCHEME

TeachingScheme		Total Credits	Examination Scheme					
(InHours)		(L+T+P/2)	Theory Marks		Practical Marks		Total	
L	Т	Р	С	СА	ESE	СА	ESE	Marks
3	-	2	4	30*	70	25	25	150

(*):Out of 30marks under the theory CA,10marks are for assessment of the micro-project to facilitate the integration of COs, and the remaining 20 marks are the average of 2 tests to be taken during the semester for assessing the attainment of the cognitive domain UOs required for the attainment of theCOs.

Legends: L-Lecture; T – Tutorial/Teacher Guided Theory Practice; P - Practical; C – Credit, CA - Continuous Assessment ;**ESE**-End Semester Examination.

5. SUGGESTEDPRACTICALEXERCISES

The following practical outcomes(PrOs) arethesubcomponents of the COs. *These PrOs need to be attained to achieve the COs.*

Sr. No.	o. Practical Outcomes (PrOs)		Approx. Hrs. Required
1	Standardize Sulfuric Acid Solution	I	2
2	Preparation of Hydrated Lime	I	2
3	Preparation of Caustic Soda	I	2
4	Preparation of Potassium Chloride	I	2
5	Preparation of Phenol Formaldehyde	II	2
6	Find out Acid Value of Oil		2
7	Preparation of Vegetable Oil from Seed		2
8	8 Preparation of Soap		2
9	Preparation of Detergent Powder		2
10	Preparation of Alcohol		2
11	Find out moisture, volatile matter and ash content in fuel	IV	2
12	12 Determine Calorific Value of Fuel		2
13	Preparation of Aspirin		2
14	Prepare of Nitrobenzene	V	2

<u>Note</u>

- *i.* More *Practical Exercises* can be designed and offered by the respective course teacher todevelop the industry-relevant skills/outcomes to match the COs. The above table is only a suggestive list. Course teacher can select any 14 practicals.
- i. The following are some sample'Process' and 'Product' related skills (more may be added/deleted depending on the course) with approximate percentage weightage that occur in the above listed **Practical Exercises** of this **cours**e required which are embedded in the Cos and ultimately the competency.

Sr.No	SamplePerformanceIndicatorsforthePrOs	Weightage in %(Approximate)
1	Prepare experimental set up accurately.	10

2	Use apparatus for precise measurements.	20
3	Practice and adapt good and safe measuring techniques.	10
4	Good Record keeping of the observations accurately.	20
5	Interpret the results and theirconclusion.	20
6	Prepare Report in prescribed format	10
7	Viva-Voce	10
	Total	100

6. MAJOR EQUIPMENT/INSTRUMENTS AND SOFTWARE REQUIRED

These major equipment/instruments and Software required to develop PrOs are given below with broad specifications to facilitate procuremen to them by the administrators/management of the institutes. This will ensure the conduction of practice in All institutions across the state in a proper way so that the desired skills are developed in students.

Sr. No.	EquipmentNamewithBroadSpecifications	PrO.No.
1	HotAirOven: Temperature is controlled by digital temperature indicator cumcontroller from ambient to 250°C with ± 0.1°C Accuracy. Powersupply: 220/230V, 50Hz single phase, Capacity (Approx.): 50 – 100 liter, Type of Shelves: 03, Material of Inner Chambers: SS304, Material of Outer Chamber: MS with powder coated paint,MaterialofShelves:SS wire mesh.	All
2	LaboratoryWeighingBalance: Type of Laboratory Balance: Analytical, Sensitivity (mg): 1 mg,Maximum Capacity of weighing (grams): 200 g, Shape of PAN:Circular, PowerSupply: Single Phase,Display: LED.	All
3	HotPlateWithMagneticStirrer: NumberofstirringPositions:1,Calibration:AutomaticCalibration,Magneti c stirrer with a hot plate, Speed Control Accuracy of setspeed (+/-) (RPM): 5, Maximum Stirring capacity per position:3000ml,TopplateMaterial:Stainlesssteel	All
4	Lab cooling bath: 220V/50HZ, 1.5KW, 370*340*480mm	All
5	Bomb calorimeter Model CC01/M3,, Iso-Thermal, BS 1016: Part 5:1967 IS: 1359–1959 IP 12/63T	12
6	Grinder: 230V 50Hz, 950 W, 11000 rpm, 1.8 K	07
7	Oil making machine : 3-6 Kg/Hr, 600 W, Gear Box, 400x160x360mm	07
8	Hand blender: 200w	09
9	Crucible and designator : white ceramic melting crucible , Dish cup 55mm for high temperature refining,	11
10	Furnace : Digital Muffle Furnace, 220-230V, 900°C, 25x125x250mm	11

11	Fermentator: 22 x 40 x 38 cm (W x D x H), LDC 4 x 40 digits with	
	backlight, Pyrex glass with 5 to 8 side necks (culture volumes from 35	11
	ml to 6 l)	

7. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of theabove-mentioned COs and PrOs. More could be added to fulfill the development of thiscompetency.

- a) Work as a leader
- b) Follow ethical practices
- c) Observe safety measures
- d) Good house keeping
- e) Time management
- f) Practice environmentally friendly methods and
- g) processes.(Environment-related)

The ADOs are best developed through laboratory/field-based exercises.Moreover,the levelof achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increaseas planned below:

- i. 'ValuingLevel'in1styear
- ii. 'Organization Level'in2ndyear.
- iii. 'Characterization Level'in3rdyear.

8. UNDERPINNINGTHEORY

The major underpinning theory is given below based on the higher level UOs of *Revised Bloom'staxonomy* that are formulated for development of theCOs and competency. If required, more such higher-level UOs could be included by the course teacher to focus on the attainment of COs and competency.

Unit	Unit Outcomes (UOs)	Topics and Sub-topics
Unit – I	1a. Classify Various Chemical Industries	1.1 Classification of Chemical Industries
Inorganic Chemical Industries	1b. Describe Properties & Uses of Chemicals	1.2 Physical Properties, Application, Manufacturing Process and Major Engineering Problem of
	1c. Prepare Flow Diagram and Explain Manufacturing Process	1.2.1 Sulphuric acid
	1d. Explain Major Engineering Problems	1.2.2 Soda ash
		1.2.3 Caustic soda
		1.2.4 Cement
		1.2.5 Lime
		1.2.6 Urea
		1.2.7 Elemental phosphorus
		1.2.8 Potassium Chloride
Unit-II	2a. Classify Polymer	2.1 Classification of Polymer
Polymer Industries	2b. Explain application & uses of polymer	2.2 Explain Physical Properties, Application & Manufacturing Process of
	2c. Prepare Flow Diagram and Explain Manufacturing Process	2.2.1 Polyethylene
		2.2.2 Styrene butadiene rubber

		2.2.3 Phenol formaldehyde
Unit -III	3a. Define fat and oil, carbohydrates, pulp	3.1 Definition and Physical Properties of Fat &
Natural	& paper	Oil
Product Industries	3b. Describe physical properties of oil & fat	3.2 Manufacturing Process of Vegetable oil
	3c. Describe fermentation types	3.3 Basics of Carbohydrates
	3d. Prepare Flow Diagram and Explain	3.5 Manufacturing Process of Sugar with
	Manufacturing Process	Major Engineering Problem
	3e. Explain Major Engineering Problems	3.6 Basics of Soap & Detergent
	3f. Explain Role of Biotechnology	3.7 Manufacturing Process of
		3.7.1 Soap
		3.7.2 Detergent Powder
		3.8 Definition of Pulp & Paper
		3.9 Manufacturing Process of Pulp by Kraft
		Process with Major Engineering Problem
		3.10 Manufacturing Process of Paper by Wet
		Process with Major Engineering Problem
		3.11 Types of Fermentation
		3.12 Manufacturing Process of Ethanol with
		Major Engineering Problem 3.13 Role of Biotechnology in Chemical
		Engineering
Unit –IV	4a. Classify, describe and uses of fuels	4.1 Fuels: types, sources, uses
Fuel and	4b. Classify Coal	4.2 Classification of Coal
Industrial	4c. Prepare Flow Diagram and Explain	
Gases	Manufacturing Process	4.3 Important industrial gases
	4d. Explain Major Engineering Problems	4.4 Manufacturing and major engineering problem of
	4e. Describe Explosive and propellant	4.4.1 Producer gas
	4f. Explain important of cryogenic	4.4.2 Coke oven gas
	technology in chemical engineering	4.4.2 Core over gas
		4.5 Classify Explosive and propellant
		4.6 Important of cryogenic technology in
ال الخامرا ا	En Classify Dharmanautical Drugs	chemical engineering
Unit –V Synthetic	5a. Classify Pharmaceutical Drugs, Pesticides and Dyes	5.1 Classification of pharmaceutical Drugs based on Their Uses with Examples
Organic		5.2 Manufacturing Process and Major
Chemical	5b. Explain pesticides formulation	Engineering Problem of
Industries	5c. Prepare Flow Diagram and Explain	5.2.1 Penicilline
	Manufacturing Process	
	5d. Explain Major Engineering Problems	5.2.2 Aspirin
		5.3 Classify pesticides
		5.4 Pesticide formulation
		5.5 Manufacturing Process and Major
		Engineering Problem of
		5.5.1 Parathion
		5.5.2 2-4 Dichlorophenoxy acetic acid

	5.6 Classification of Dyes
	5.7 Manufacturing Process of Nitrobenzene
	5.8 Manufacturing Process of Aniline by
	Reduction of Nitrobenzene

9. SUGGESTEDSPECIFICATIONTABLEFORQUESTIONPAPERDESIGN

		D	Dist	stribution of TheoryMarks			
UnitNo.	UnitTitle	TeachingHo urs	R Level	U Level	A Level	TotalM arks	
I	Inorganic Chemical Industries	09	04	07	04	15	
II	Polymer Industries	05	02	03	03	08	
	Natural Product Industries	14	06	10	06	22	
IV	Fuel and industrial gases	05	02	04	04	10	
V	Synthetic Organic Chemical Industries	09	03	06	06	15	
Total		42	17	30	23	70	

Legends:R=Remember,U=Understand,A=Apply and above(Revised Bloom'staxonomy)

Note: This specification table provides general guidelines to assist students for their learning and to teachers to teach and question paper designers/setters to formulate test items/questionsto assess the attainment of the UOs. The actual distribution of marks at different taxonomy levels(of R, Uand A)in the question paper may slightly vary from above table.

10. SUGGESTED STUDENT ACTIVITIES

Other than the classroom and laboratory learning, the following are the suggested studentrelated *co-curricular* activities which can be undertaken to accelerate the attainment of the various out comes in this course: Students should perform the following activities in group and prepare small reports of about 5 pages for each activity. They should also collect/record physical evidence such as photographs/videos of the activities for their (student's) portfolio which will be useful fortheir placement interviews:

- a) Prepare a PowerPoint presentation or animation showing different types of chemical manufacturing Process
- b) Prepare a model of different chemical product flow diagram
- c) Preparation of a table showing the difference between Organic and Inorganic Compounds.
- d) Market survey of different Chemical product and compare their physical and chemical properties.
- e) Library survey regarding polymers and fertilizers in different industries.
- f) Collect different polymers and prepare the chart/ PowerPoint based on their type, properties, and uses.

11. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES(ifany)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Massive open online courses (*MOOCs*) may be used to teach various topics/subtopics.
- b) Guide student(s)inundertakingmicro-projects/activities.
- c) Different types of teaching methods i.e. video demonstration, activity based learning, casestudy, m-learning need to be employed byteachers to develop the outcomes.
- d) Some *of the topics/sub-topics* which are relatively simpler or descriptive are to begiven to the students for *self-learning* but to be assessed using different assessment methods.
- e) Teachers need to ensure to create opportunities and provisions forco-curricular activities.
- f) Guide students to address issues on environment and sustainability with reference to using the knowledge of this course
- g) OERs, Vlab, and Olabs may be used to teach for the teaching of different concepts.

12. SUGGESTEDMICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/heart the beginning of the semester. In the first four semesters, themicro-projectare group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

Themicro-projectcould beindustry application-based, internet-based, workshopbased, laboratory-based, or field-based. Each micro-project should encompass two or moreCOs which are the integration of PrOs, UOs, and ADOs. Each student will have to maintain a dated work diary consisting of individual contributions in the project work and give a seminar presentation of it before submission. The duration of the microproject should be about **14-16** *(fourteen to sixteen) student engagement hours* during the course.The student sought to submit micro-project by the end of the semester (so that they develop industry-orientedCOs.

A suggestive listofmicro-projects is givenhere. This should relate highly to the competency of the course and theCOs.Similar micro-projects could be added by the concerned course teacher:

- a) Prepare a chart of the properties of given product
- b) Prepare a chart to demonstrate manufacturing process.
- c) Prepare a report on major engineering problem of given manufacturing process
- d) Prepare a chart of application of given products
- e) Prepare a power pont presentation on a topic "List of chemicals manufacturing industries in India"
- f) Prepare a PowerPoint presentation or animation showing different types of chemical manufacturing Process

13. SUGGESTEDLEARNINGRESOURCES

Sr.	Title ofBook	Author	Publication with the
No.		Autio	place,year,and ISBN

1	Outlines of Chemical	M. Gopala Rao,	Affiliated East West Press
	Technology, 3rd edition	Marshall Sittig	(Pvt) Ltd-New Delhi
2	Shreve's Chemical Process Industries, 5th edition	Austin G.T.	McGraw Hill publication – New Delhi
3	Chemical Technology –	G.N. Pandey and	Vani Books Company
	Vol. I and II, 2nd edition	Shukla	-Hyderabad
4	A Text Book on Petrochemicals, 2nd edition	Rao B. K. B.	Khanna Publishers –New Delhi

14. SUGGESTEDLEARNINGWEBSITES

- 1. http://www.epa.gov/sectors/sectorinfo/sectorprofiles/chemical.html
- 2. www.emis.vito.be/sites/default/Bref_cement_and_lime_production.pdf
- 3. www.docbrown.info/page04/Mextract.htm
- 4. http://www.contentshoppe.com/images/eLearning/sample2.swf
- 5. http://www.auroma.in/propertiescoal.pdf
- 6. www.naturalproductsexpoindia.com/
- 7. www.andritz.com/pulp-and-paper/pp-pulp-production.htm
- 8. ewww.linde-gas.com/en/products_and_supply/gases_fuel/index.htm
- 9. www.iisrp.com/WebPolymers/00Rubber_Intro.pdf
- 10. http://www.niehs.nih.gov/health/topics/agents/pesticides/

PO-COMPETENCY-COMAPPING

Semester III	Chemical Process Technology (CourseCode: 4330504)								
Semester m		Program Outcomes							
	PO1	PO2	PO3	PO4	PO5	PO6	PO7		
Competency &CourseOutcomes	Discipli ne-	mAnal ysis		ngTools,E xperimen	gpractices forsociety,s ustainabilit		Life- longlearni ng		
Competency UseSyn	-								
thesize reactions									
and unit operations									
steps to develop and									
operate a chemical	3	2	-	2	2	2	1		
plant to									
manufacture									

important chemicals							
CourseOutcomes CO1:Explain the classification and							
properties of various chemicals	3	-	-	2	-	2	1
CO2:Apply concept of the							
manufacturing processes of various chemicals with neat sketch to operate chemical plant.	3	2	-	3	2	2	2
CO3:Identify major engineering							
problems encountered in manufacturing processes.	2	2	-	-	2	-	-
CO4:Suggest applications of							
various chemicals.	2	-	-	1	1	-	1

Legend: '**3**' forhigh, '**2**' formedium, '**1**' forlowand'-' forno correlationofeachCOwithPO.

COURSE CURRICULUM DEVELOPMENT COMMITTEEGTUResourcePersons

S. No.	Nameand Designation	Institute	ContactNo.	Email
1.	Mr. R P Hadiya	G P Rajkot		rphadiya@yahoo.co.in
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	Engg			
2.	Ms. C. B. Desai	Government		desaicharmib@gmail.com
	Lecturer in Chemical	Polytechnic		
	Engineering	Gandhinagar		

GUJARAT TECHNOLOGICAL UNIVERSITY (GTU)

Competency-focused Outcome-based Green Curriculum-2021 (COGC-2021) Semester-V

Course Title: Entrepreneurship & Start-ups

(Course Code: 4300021)

Diploma programmer in which this course is offered	Semester in which offered
All Branches of Diploma Engineering	5 th Semester

1. RATIONALE

Entrepreneurs have significant impact on our country's current developing economy. The social expectations towards engineering professionals are certainly emerging as job creators especially with the thrust given to "Make in India" and "Vocal for Local" campaigns. Startup India is a well-known flagship initiative of the Government of India, intended to catalyze startup culture and build a strong and inclusive ecosystem for innovation and entrepreneurship. The last 6 years have witnessed tremendous growth of start-ups i.e. from 733 in 2016-17 to 14000 in 2021-22. This course focuses on the basic roles, skills and functions of entrepreneurship with special attention to startup. The course is directed to help students to enhance capabilities in the field of managing the given task as well as to understand peripheral influencing aspects for starting a new business. It will certainly help students to think in a direction to establish a small industry /start-up and develop /validate it using fundamental know how.

2. COMPETENCY

The purpose of this course is to help the student to attain the following industry identified competency through various teaching learning experiences:

• Establish a small enterprise /start-up validate it and make it scalable.

3. COURSE OUTCOMES (COs)

The practical exercises, the underpinning knowledge and the relevant soft skills associated with the identified competency are to be developed in the student for the achievement of the following COs:

Upon completion of the course, the student will be able to demonstrate knowledge of the following topics:

- 1) Understanding the dynamic role of entrepreneurship and Startups by Acquiring Entrepreneurial spirit and resourcefulness, quality, competency, and motivation
- 2) Identify a Business Idea and implement it
- 3) Select suitable Management practices like leadership and Ownership, resource institutes
- 4) Overview of Support Agencies and Incubators
- 5) Building Project Proposal & knowing CSR , Ethics, Ex-Im, & Exit strategies

4. TEACHING AND EXAMINATION SCHEME

Teaching Scheme	Total Credits	Examination Scheme					
(In Hours)	(L+T+P/2)	Theory Marks	Practical	Total Marks			

						Marks		
L	Т	Р	С	СА	ESE	СА	ESE	
3	0	0	3	30*	70	0	0	100

(*): Out of 30 marks under the theory CA, 10 marks are for assessment of the micro-project to facilitate integration of COs and the remaining 20 marks is the average of 2 tests to be taken during the semester for the assessing the attainment of the cognitive domain UOs required for the attainment of the COs.

Legends: L-Lecture; T – Tutorial/Teacher Guided Theory Practice; P - Practical; C – Credit, CA - Continuous Assessment; ESE - End Semester Examination.

5. SUGGESTED Soft PRACTICAL EXERCISES (During Theory)

The entrepreneurial or start-up journey begins by readying for your future dream from college projects and pursuing the same beyond college hours also. It is encouraged to go through COs and identify traits and search for various state and national agencies for your entrepreneurship / start-up journey and convert the same into successful product in market.

The following practical outcomes (SPrOs) are the sub-components of the Course Outcomes (COs). Some of the **SPrOs** marked **'*'** are compulsory, as they are crucial for that particular CO at the 'Precision Level' of Dave's Taxonomy related to 'Psychomotor Domain'.

<u>Note</u>

- *i.* Though the course does not contain any Practical work, a few **Practical Exercises** can be designed and offered by the respective course teacher to develop the industry relevant skills/outcomes to match the COs. The below table is only a suggestive list.
- ii. The following are some **sample** 'Process' and 'Product' related skills (more may be added/deleted depending on the course) that occur in the above listed **Practical Exercises** of this course required which are embedded in the COs and ultimately the competency.

Sr. No.	Sample Performance Indicators for the PrOs	Weightage in %
1	Entrepreneur Traits and Behavior Modelling	30
2	Various State and Central Entrepreneurship	30
	Promotional Schemes and Start-up Policies	
3	Business Model for a Startup and study of Unicorns*	40
	Total	100

6. AFFECTIVE DOMAIN OUTCOMES

The following *sample* Affective Domain Outcomes (ADOs) are embedded in many of the abovementioned COs and PrOs. More could be added to fulfill the development of this course competency.

- a. Work as a leader/a team member (while doing a micro-project).
- b. Model behavioral practices of an entrepreneur while planning for an enterprise
- c. Practice ethics and consider methods/ processes that reduce waste and/or possibly conserve environment in designing a new business till it's commercialization.

The ADOs are best developed through the laboratory/field-based exercises. Moreover, the level of achievement of the ADOs according to Krathwohl's 'Affective Domain Taxonomy' should gradually increase as planned below:

- i. 'Valuing Level' in 1st year-Planning
- ii. 'Organization Level' in 2nd year-Model Development
- iii. 'Characterization Level' in 3rd year-Make it Scalable

7. UNDERPINNING THEORY

The major underpinning theory is given below based on the higher level UOs of *Revised Bloom's taxonomy* that are formulated for development of the COs and competency. If required, more such UOs could be included by the course teacher to focus on attainment of COs and competency.

Unit	Major LearningOutcomes (in cognitive domain)		Topics and Sub-topics
to Entrepreneu rship and	 1b) Discuss characteristics and functions of entrepreneurship. 1c) Identify different types of Entrepreneurships 1d) Compare the concepts entrepreneurand intrapreneur and find out the motivation behind it 1e) Distinguish between entrepreneur and managers 1f) Identify 7-M Resources 	1. 2. 3. 4. 5. 6. 7. 8. 9.	Definition, Traits of an entrepreneur, Functions of Entrepreneurship - Job Creation, Innovation, Inspiration, Economic Development Types of Entrepreneurship Motivation for Intrapreneurship Types of Business Structures, Similarities and differences between entrepreneurs and managers. 7-M Resources Micro, Small, Medium Enterprise/ MSME - Industry Registration Process Startup India, Standup India and SSIP Gujarat & Startup registration process

	Da) Finding Ideas and making an	1	Discovering ideas and visualizing the
2) Business	2a) Finding Ideas and making an	1.	business with Activity map
Ideas and	activity map		1.1 Idea Generation
their	2 h) Develop the plane foreresting		1.2 Product Identification
implement	2b) Develop the plans forcreating	2.	Business Plan- The Marketing Plan and
ation (Idea	and starting the business		Financial Plan/ Sources of Capital
to Start-up)		3.	Business opportunity identification and
	2c) To identify business using the		evaluation
	ideation canvas and the business	4.	Market research
	model canvas		4.1.1. Questionnaire design
			4.1.2.Sampling
	2d) To know market research		4.1.3. Market survey
	related terms		4.1.4. Data analysis & interpretation
		5.	Marketing Mix (4Ps- product, price,
	2e) To know market mix related		promotion place)
	terms		5.1.1.Identifying the target market
			5.1.2. Competition evaluation and
	2f) Learn Product related		Strategy adoption
	terminologies		5.1.3. Market Segmentation
			5.1.4. Marketing, Advertising and
	2g) Emphasize on Innovation		Branding
			5.1.5. Digital Marketing
	2h) Explain concept of Risk and		5.1.6.B2B, E-commerce and GeM
	SWOT	6.	Product Terms- PLC, Mortality Curve
			and New product Development Steps,
			Inventory, Supply Chain Management
		7.	Importance and concept of Innovation, Sources and Process
		8.	Risk analysis and mitigation by SWOT
			Analysis
3) Management	3a) Explain the concept and	1.	Industry, Commerce and Business
Practices	differences between industry,	2.	Types of ownership in the
	commerce and business.		organization -Definition,
	3b) Describe various types of		Characteristics, Merits & Demerits
	ownerships in the organization.	3.	Different Leadership Models
	3c) Explain different types of	4.	Functions of Management- Merits &
	leadership models.		Demerits
	3d) Analyze the nature and		4.1 Planning
	importance of various functions of		4.2 Company's Organization Structure
	management		4.3 Directing
	3e) Discuss Financial organization		4.4 Controlling
	Management		4.5 Staffing- Recruitment and
	3f) Distinguish	_	management of talent.
	management and administration	5. c	Financial organization and management
		6.	Differences between Management
			and Administration

1) Support	4a) Identify support agencies and	1.	State & National Level Support
4) Support		1.	
Agencies and	current promotional schemes for		agencies and Current Promotional
Incubators	enterprise and startups	_	Schemes for new Enterprise
		2.	Start-up Incubation and modalities
	4b) Advocacy to investor	3.	Communication of Ideas to potential
			investors – Investor Pitch
	4c) To Explain various Legal Issues	4.	Legal Issues
			4.1. Contracts
			4.2. Copyrights
			4.3. Insurance
			4.4. IPR
			4.5. Licensing
			4.6. Patents
			4.7. Trade Secrets
			4.8. Trademarks
5) Project	5a) To work on the development	1	. Project Planning
Proposal &	of a project proposal		i. Project planning and report
Exit strategies			ii. Feasibility study
Exit Strategies	5b) Describe social responsibility		iii. Project cost estimation
	and relate with economic		iv. Breakeven point,
	Performance.		v. Return on investment and Return
			on sales
	5c) Explain managerialethics	2	
			Economic performance
	5d) To know Ex-Im Policies	3	
	Suf To know Ex III T oncies	4	
		5	•
	5e) Identify suitable strategies of	6	
	succession and harvesting	0	

8. SUGGESTED SPECIFICATION TABLE FOR QUESTIONPAPER DESIGN

Unit	Unit	Teaching	Distribution of Theory Marks				
No.			R Level	U Level	A Level	Total Marks	
I	Introduction to Entrepreneurship and Start – Ups	08	4	6	2	12	
11	Business Ideas and their implementation (Idea to Startup)		6	4	4	14	
- 111	Management Practices	12	6	8	8	22	
IV	Support Agencies and	08	4	4	4	12	

GTU - COGC-2021 Curriculum

	Incubators						
V	Project Proposal strategies	& Exit	06	2	4	4	10
	Total		42	22	26	22	70

Legends: R=Remember, U=Understand, A=Apply and above (Revised Bloom's taxonomy) <u>Note</u>: This specification table provides general guidelines to assist students for their learning and to teachers to teach and question paper designers/setters to formulate test items/questions to assess the attainment of the UOs. The actual distribution of marks at different taxonomy levels (of R, U and A) in the question paper may slightly vary from above table.

9. SUGGESTED STUDENT ACTIVITIES

Other than the classroom learning, following are the suggested student-related **co-curricular** activities which can be undertaken to accelerate the attainment of the various outcomes in this course. Students should make a portfolio i.e. perform at least FIVE from following list of activities individually or in group (not more than 2). They should prepare reports of about 2-5 pages for each activity and collect/ record physical evidence for their portfolio which may be useful for their placement interviews:

- i. Develop two products from household waste (attach photographs).
- ii. Download product development and innovative films from internet.
- iii. Prepare a collage for "Traits of successful entrepreneurs."/ "Motivation & Charms of Entrepreneurship"
- iv. Invite entrepreneurs, industry officials, bankers for interaction. Interview at least four entrepreneurs or businessman and identify
- v. Identify your hobbies and interests and convert them into business idea.
- vi. Mock Business Model- Choose a product and design a unique selling preposition, brand name, logo, advertisement (print, radio, and television), jingle, packaging, and labeling for it.
- vii. Develop your own website. Share your strengths and weakness on it. Declare your time bound goals and monitor them on the website.
- viii. Choose any product/ advertisement and analyze its good and bad points/ cost sheet/ supply chain etc. (individuals should select different ads)
- ix. Compare schemes for entrepreneurship promotion of any bank.
- x. Visit industrial exhibitions, trade fairs and observe nitty-gritty of business. Get news of Vibrant Gujarat Events. (Upcoming in Jan 2024)
- xi. Open a savings account and build your own capital.
- xii. Arrange a visit to a Mall, observe products, supply chain management and prepare report.
- xiii. Organize industrial visit and suggest modifications for process improvement. Conduct a market survey for a product /project before visit. In the visit collect data on machinery specifications, price, output/hour, power consumption, manpower requirement, wages, raw material requirement, specification, price, competitor's product price, features, dealer commissions, marketing mix etc. Make a detailed report at the end of the visit.
- xiv. Select a social cause, set objectives, plan and work for its accomplishment. Find details about some famous NGOs
- xv. Present Own Dream Start-up story as Seminar OR Analyze 2 products from Shark Tank program.

10. SUGGESTED SPECIAL INSTRUCTIONAL STRATEGIES (if any)

These are sample strategies, which the teacher can use to accelerate the attainment of the various outcomes in this course:

- a) Massive open online courses (*MOOCs*) may be used to teach various topics/subtopics.
- b) Guide student(s) in undertaking micro-projects.
- c) *'L' in section No. 4 means* different types of teaching methods that is to be employed by teachers to develop the outcomes.
- d) Show animation/video related to course content.
- e) Various Apps related to subject topics/ sub-topics
- f) Other Common instructions as under
 - 1) Instructors should emphasize more on exemplary and deductive learning.
 - 2) Students should learn to recognize, create, shape opportunities, and lead teams forproviding economic-social value to society.
 - 3) Business simulations should be used to enhance behavioral traits of successfulintrapreneurs and entrepreneurs amongst students.
 - 4) Emphasis should be on creating entrepreneurial society rather than only setting upof enterprise.
 - 5) They must be encouraged to surf on net and collect as much information aspossible.
 - 6) Each student should complete minimum ten activities from the suggested list. Minimum possible guidance should be given for the suggested activities.
 - 7) Students should be promoted to use creative ideas, pool their own resources, finishtheir presentation, communication and team skills.
 - 8) Alumni should be frequently invited for experience sharing, guiding and rewarding students.
 - 9) Display must be arranged for models, collages, business plans and othercontributions so that they motivate others.
 - 10) You may show video/animation film / presentation slides to demonstrate variousmanagement functions, traits of entrepreneur etc.
 - 11)Arrange a visit to nearby venture capital firm.
 - 12) Give 1 Mini project and 1 project report for future business to all the students.
 - 13) The following pedagogical tools will be used to teach this course:
 - a) Lectures and Discussions
 - b) Role Playing
 - c) Assignments and Presentations
 - d) Case Analysis
 - e) Quiz on Management and Entrepreneurship
 - g) Mimic/ narrate examples from world's leading businessmen among the students.
 - h) Guide students on how to address issues on environment and sustainability

12. SUGGESTED MICRO-PROJECTS

Only one micro-project is planned to be undertaken by a student that needs to be assigned to him/her in the beginning of the semester. In the first four semesters, the micro-projects are group-based (group of 3 to 5). However, **in the fifth and sixth semesters**, the number of students in the group should **not exceed three**.

The micro-project could be industry application based, internet-based, workshop-based, laboratorybased or field-based. Each micro-project should encompass two or more COs which are in fact, an integration of PrOs, UOs and ADOs. Each student will have to maintain a dated work diary consisting of individual contributions in the project work and give a seminar presentation of it before submission. The duration of the guidance for micro project should be about **6-8** (*six to eight*) *student engagement hours* during the theory/ course. The students ought to submit micro-project by the end of the semester to develop the industry-oriented COs.

A suggestive list of micro-projects/ practical exercise is given here. This has to match the competency and the COs. Similar micro-projects could be added by the concerned course teacher:

(It can be a Seminar with bound /hand written notes/ ppts of individual students OR a product/ service portfolio)

- 1) Entrepreneur Traits and Behavior Modelling
- 2) Various State and Central Entrepreneurship Promotional Schemes and Start-up Policies
- 3) Business Model for a Startup and study of Unicorns
- 4) Make your own Product / Service portfolio/ Proposal with USP, logo, advertisement (print, radio, and television), jingle, packaging, labeling and branding for it.

Sr. No	Title of Book	Author	Publication with place, year and ISBN		
1	Entrepreneurship in Action	Coulter PHI 2nd Edition			
2	Entrepreneurship Development	E. Gordon & K. Natarajan	Himalaya		
3	Entrepreneurship	Robert D. Hisrich & Mathew J. Manimala	McGraw Hill Education; ISBN 978-1259001635		
4	Entrepreneurial Development	S S Khanka	S Chand & Company; ISBN: 978-8121918015		
5	Entrepreneurship Development and Management	A. K. Singh	Jain Book Agency (JBA) publishes, New Delhi		
6	Entrepreneurship Development& Management	R.K. Singal	S K Kataria and Sons; ISBN: 978-8189757007		
7	Small Scale Industries and Entrepreneurship	Vasant Desai	Himalaya 2008		
8	Entrepreneurship	Roy Rajeev	Oxford University Press; ISBN: 978-0198072638		
9	Industrial Engineering and Management	O.P.Khanna	Dhanpat Rai and Sons, Delhi		
10	Industrial Organization and Management	Tara Chand	NemChand and Brothers; Roorkee		
11	Industrial Management and Entrepreneurship	V. K. Sharma.	Scientific Publishers, New Delhi		
12	Entrepreneurship Developmentand Small Business Enterprise	Poornima M Charantimath	Pearson Education; ISBN: 978-8131759196		
13	Entrepreneurship Development	S Anil kumar	NEW AGE Intern. Pvt Ltd; ISBN: 978-8122414349		

13. SUGGESTED LEARNING RESOURCES

r	1	1	1
14	The Startup Owner's	Steve Blank and Bob	K & S Ranch
	Manual: The Step-by-Step	Dorf	ISBN – 978-0984999392
	Guide for Building a Great		
	Company		
15	The Lean Startup: How	Eric Ries	Penguin UK
	Today's Entrepreneurs Use		ISBN - 978-0670921607
	Continuous Innovation to		
	Create Radically Successful		
	Businesses		
16	Demand: Creating What	Adrian J. Slywotzky with	Headline Book Publishing
	People Love Before They	Karl Weber	ISBN – 978-0755388974
	Know They Want It		
17	The Innovator's Dilemma: The	Clayton M. Christensen	Harvardbusiness ISBN: 978-
	Revolutionary Book That Will		142219602
	Change the Way You Do		
	Business		
18	How to write a business plan,	Brian Finch	2nd edition, 2007, Kogan
			Page India Pvt. Ltd.
	Advance Reading		
19	HBR - Creating business plan	-	20-minute manager series,
			2014.
20	HBR – Creating business plan	-	Expert solution to everyday
			challenges, 2007.

14. SOFTWARE/LEARNING WEBSITES

[A] List of Software/Learning Websites:

Sr. No.	Topic Key Word	Link		
1	MoCl	<u>https://www.india.gov.in/website-ministry-</u> <u>commerce-and-industry</u>		
2	MSME	1)https://msme.gov.in/ 2) https://www.msmex.in/learn/government- schemes-for-startups-and-msmes-in-india/		
3	Start-up, Stand-up India & SSIP Gujarat	 <u>1) https://www.startupindia.gov.in/</u> <u>2) https://www.standupmitra.in</u> <u>3) https://udyamimitra.in/page/standup-india-loans</u> <u>4) https://www.ssipgujarat.in/</u> 		
4	Make in India	https://www.makeinindia.com/		
5	Atmanirbhar Bharat Abhiyan Vocal for Local	https://indiancc.mygov.in > uploads > 2021/08		
6	Skill India	https://skillindia.gov.in		
7	MSDE	https://www.msde.gov.in/		
8	Vibrant Gujarat	https://www.vibrantgujarat.com/		
9	NABARD	www.nabard.com		
10	PAN	https://www.onlineservices.nsdl.com/paam/endUs erRegisterContact.html		
11	l-hub	https://ihubgujarat.in		
12	GSTIN	https://reg.gst.gov.in/registration		

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13	IEC Code	https://www.dgft.gov.in/CP		
14	Mudra	https://www.mudra.org.in/		
15	Export-Import	http://niryatbandhu.iift.ac.in/exim/		
16	NSIC	https://www.nsic.co.in/		
17	DIC	<u>https://ic.gujarat.gov.in/dic -contact.aspx -District</u> Industries Centre		
18	EDI	https://www.ediindia.org/		
19	CED	https://ced.gujarat.gov.in/home		
20	NIESBUD	https://www.niesbud.nic.in/		
21	Start-up Talky	https://startuptalky.com/list-of-government- initiatives-for-startups/		
22	Invest India	https://www.investindia.gov.in/startup-india-hub		
23	SAAC	https://www.saccindia.org/india/startups.html?utm source=google&utm_medium=cpc&gclid=EAIaIQo bChMIutLQ4dfW_wIVeppmAh1cOAAIEAMYASAAEgI JO_D_BwE		
24	Action for India	https://actionforindia.org/afi-activity-accelerator- programs.html?gclid=EAIaIQobChMIutLQ4dfW_wIV eppmAh1cOAAIEAMYAiAAEgLVGvD_BwE		
25	Indian Chamber of Commerce	https://www.indianchamber.org/		
26	FICCI	https://www.ficci.in/api/home		
27	GCCI	https://www.gujaratchamber.org/		

[B] Some Films (To be seen on Sundays/holidays by students on their own, not to be shown in polytechnics in any case**)**

- i. Any Body Can Dance (2013)
- ii. Corporate (2006)
- iii. Do Duni Char (2010)
- iv. Guru (2007)
- v. Oh My God (2013)
- vi. Pirates of Silicon Valley (1999)
- vii. The Pursuit of Happiness (2006)
- viii. Rocket Singh (2010)
 - ix. Start-up.com (2001)
 - x. The Social Network (2010)
- xi. Wall Street (1987)
- xii. Band Baja Barat (2010)
- xiii. You've Got Mail (1998)
- xiv. Steve Jobs (2015)
- xv. Chef (2014)
- xvi. "Office Space (1999)
- xvii. Erin Brockovich (2000)
- xviii. The Founder (2016)

15. PO-COMPETENCY-CO MAPPING:

Semester	V
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	(Course Code: 4300021)						
	POs						
Competency & Course Outcomes	PO 1 Basic & Discipline specific knowledge	PO 2 Problem Analysis	PO 3 Design / develo pment of solutio n	PO4 Engineering Tools, Experiment ation & Testing	PO 5 Engineering practices for society, sustainability & environment	PO 6 Project Managem ent	PO 7 Life-long learning
Competency	Use co	ncepts o	of mana	gement op	otimally to e	establish	a small
<u>Competency</u>	enterp	rise or s	tart-up,	validate it	t and make	it scalab	le.
CO1-Understanding the dynamic role of entrepreneurship and Startups by Acquiring Entrepreneurial spirit and resourcefulness, quality, competency, and motivation	3	1	2	-	-	2	2
CO2- Identify a Business Idea and implement it	3	2	2	1	1	3	3
CO3-Select suitable Management practices like leadership and Ownership, resource institutes	3	-	1	1	2	2	3
CO4- Overview of Support Agencies and Incubators	2	3	2	2	1	2	2
CO5- Building Project Proposal & knowing CSR, Ethics, Ex-Im, & Exit strategies	3	2	2W	1	1	3	3

Legend: '**3'** for high, '**2**' for medium, '**1'** for low and '-' for no correlation of each CO with PO.

16. COURSE CURRICULUM DEVELOPMENT COMMITTEE

GTU Resource Persons

S. No	Name and Designation	Institute	Contact No.	Email
1.	Mr. Ujjval V Buch (MBA)	G.P.Ahmedabad	9825346922	uvbuch@gmail.com
2.	Dr. Satya Acharya	EDI, Bhat.	7600050606	<u>satya@ediindia.org</u>